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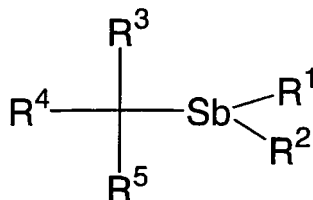
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(54) **ORGANIC ANTIMONY COMPOUND, PROCESS FOR PRODUCING THE SAME, LIVING RADICAL POLYMERIZATION INITIATOR, PROCESS FOR PRODUCING POLYMER USING THE SAME, AND POLYMER**

(57) 1. An organoantimony compound represented by the formula (1), processes for producing polymers with use of the compound, and polymers



(1)

wherein R<sup>1</sup> and R<sup>2</sup> are C<sub>1</sub>-C<sub>8</sub> alkyl, aryl, substituted aryl or an aromatic heterocyclic group, R<sup>3</sup> and R<sup>4</sup> are each a hydrogen atom or C<sub>1</sub>-C<sub>8</sub> alkyl, and R<sup>5</sup> is aryl, substituted aryl, an aromatic heterocyclic group, oxycarbonyl or cyano. P

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## Description

## TECHNICAL FIELD

5 **[0001]** The present invention relates to organoantimony compounds and a process for preparing the same. More particularly, the invention relates to living radical polymerization initiators of the organoantimony type, a process for preparing living radical polymers using the initiator and living radical polymers. The present invention also relates to a process for preparing a random copolymer and the random copolymer, a process for preparing a block copolymer and the block copolymer, and these macro living radical polymerization initiators and polymers.

10 **[0002]** Further, the polymer of the present invention is suitable for use as a resist material and the like usable for preparing a semiconductor device.

## BACKGROUND ART

15 **[0003]** Living radical polymerization is a polymerization process which is adapted for precision control of molecular structures while ensuring convenience and universal usefulness of radical polymerization, and is powerful means for preparing novel high polymer materials. The present inventors have reported, as an example, a living radical polymerization using an organotellurium compound as an initiator (for example, patent literature 1).

20 [patent literature 1] WO 2004/14848

**[0004]** The process of patent literature 1 makes it possible to control molecular weights and molecular weight distributions, but utilizes an organotellurium initiator and has no disclosure about organoantimony compounds of the present invention.

25 **[0005]** An object of the present invention is to provide a process for preparing living radical polymers and the polymers, which makes possible precision control of molecular weights and molecular weight distributions ( $PD = Mw/Mn$ ), by polymerizing a vinyl monomer using an organoantimony compound.

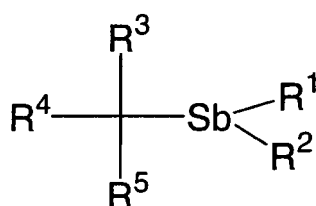
30 **[0006]** An object of the present invention is to provide a process for preparing living radical polymers and the polymers, which makes possible precision control of molecular weights and molecular weight distributions ( $PD = Mw/Mn$ ) under mild conditions, in short period of time and in high yield for preparing the compound, a process for producing a polymer with use of the compound, and the polymer.

## DISCLOSURE OF THE INVENTION

35 **[0007]**

1. An organoantimony compound represented by the formula (1)

40



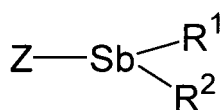
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(1)

50 wherein  $R^1$  and  $R^2$  are  $C_1$ - $C_8$  alkyl, aryl, substituted aryl or an aromatic heterocyclic group,  $R^3$  and  $R^4$  are each a hydrogen atom or  $C_1$ - $C_8$  alkyl, and  $R^5$  is aryl, substituted aryl, an aromatic heterocyclic group, oxycarbonyl or cyano.

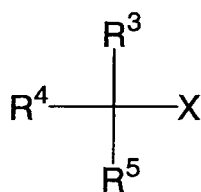
2. A process for preparing an organoantimony compound of the formula (1) comprising reacting a compound of the formula (3), and a compound of the formula (4) or (5)

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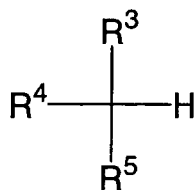
(3)

(wherein R<sup>1</sup> and R<sup>2</sup> are as defined above, and Z is a halogen atom or alkali metal)



(4)

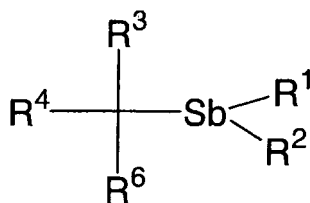
(wherein R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as defined above, and X is a halogen atom)



(5)

(wherein R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as defined above).

3. A living radical polymerization initiator of the formula (2)



(2)

wherein R<sup>1</sup> and R<sup>2</sup> are C<sub>1</sub>-C<sub>8</sub> alkyl, aryl, substituted aryl or an aromatic heterocyclic group, R<sup>3</sup> and R<sup>4</sup> are each a hydrogen atom or C<sub>1</sub>-C<sub>8</sub> alkyl, and R<sup>6</sup> is aryl, substituted aryl, an aromatic heterocyclic group, acyl, oxycarbonyl or cyano.

4. A process for producing a living radical polymer characterized in that a vinyl monomer is polymerized with use of a living radical polymerization initiator of the formula (2).

5. A process for producing a living radical polymer characterized in that a vinyl monomer is polymerized with use of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.

6. A process for producing a random copolymer characterized in that at least two vinyl monomers are polymerized with use of a living radical polymerization initiator of the formula (2), or this initiator and an azo type polymerization initiator.

7. A macro living radical polymerization initiator which is obtainable by subjecting a vinyl monomer to polymerization with use of a living radical polymerization initiator of the formula (2), or this initiator and an azo type polymerization initiator.

8. A process for producing a living radical polymer characterized in that a vinyl monomer is polymerized with use of a living radical polymerization initiator of the formula (2), at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound, and as required an azo type polymerization initiator.

9. A process for producing a resin containing an acid-dissociable group characterized in that a vinyl monomer is polymerized with use of one of the following (a) to (d),

(a) a living radical polymerization initiator of the formula (2),

(b) a mixture of a living radical polymerization initiator of the formula (2), and an azo type polymerization initiator,

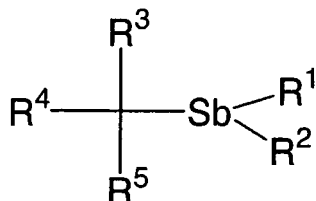
(c) a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound, and

(d) a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine com-

pound.

10. A radiation-sensitive resin composition comprising a resin containing an acid-dissociable group, and a radiation-sensitive acid producing agent.

[0008] The organoantimony compounds of the present invention are represented by the formula (1)



(1)

wherein  $R^1$  and  $R^2$  are  $C_1$ - $C_8$  alkyl, aryl, substituted aryl or an aromatic heterocyclic group,  $R^3$  and  $R^4$  are each a hydrogen atom or  $C_1$ - $C_8$  alkyl, and  $R^5$  is aryl, substituted aryl, an aromatic heterocyclic group, oxycarbonyl or cyano.

[0009] Examples of groups represented by  $R^1$  and  $R^2$  are as follows.

[0010] Examples of  $C_1$ - $C_8$  alkyl groups usable are straight-chain, branched chain or cyclic alkyl groups having 1 to 8 carbon atoms, such as methyl, ethyl, n-propyl, isopropyl, cyclopropyl, n-butyl, sec-butyl, tert-butyl, cyclobutyl, n-pentyl, n-hexyl, n-heptyl and n-octyl. Preferable alkyl groups are straight-chain or branched chain alkyl groups having 1 to 4 carbon atoms. Methyl, ethyl or n-butyl is more preferable.

[0011] Examples of aryl groups are phenyl and naphthyl. Preferable is phenyl. Examples of substituted aryl groups are phenyl having a substituent and naphthyl having a substituent.

[0012] Examples of substituents of aryl groups having a substituent are a halogen atom, hydroxyl, alkoxy, amino, nitro, cyano, carbonyl-containing groups represented by  $-COR^a$  ( $R^a = C_1$ - $C_8$  alkyl, aryl,  $C_1$ - $C_8$  alkoxy or aryloxy), sulfonyl, trifluoromethyl, etc. Preferable aryl group having a substituent is trifluoromethyl-substituted phenyl. Preferably such substituted groups have one or two substituents at the para-position or ortho-position. Examples of aromatic heterocyclic groups are pyridyl, pyrrol, furyl and thienyl.

[0013] Examples of groups represented by  $R^3$  and  $R^9$  are as follows.

[0014] Examples of  $C_1$ - $C_8$  alkyl groups usable are the same as the alkyl groups represented by  $R^1$  and given above.

[0015] Examples of groups represented by  $R^5$  are as follows.

[0016] Examples of aryl, substituted aryl, aromatic heterocyclic groups usable are the same as those groups represented by  $R^1$  and given above.

[0017] Examples of preferred oxycarbonyl groups are those represented by  $-COOR^b$  ( $R^b = H, C_1$ - $C_8$  alkyl or aryl) such as carboxyl, methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, n-butoxycarbonyl, sec-butoxycarbonyl, ter-butoxycarbonyl, n-pentoxycarbonyl and phenoxycarbonyl. Methoxycarbonyl and ethoxycarbonyl are more desirable oxycarbonyl groups.

[0018] Examples of preferred groups represented by  $R^5$  are aryl, substituted aryl, oxycarbonyl and cyano. The aryl group is preferably phenyl. Examples of preferred substituted aryl groups are phenyl substituted with a halogen atom and phenyl substituted with trifluoromethyl. When the substituent is a halogen, the phenyl is substituted with preferably one to five halogen atoms. In the case of alkoxy or trifluoromethyl, preferably one or two substituents are present. When having one substituent, the group is substituted preferably at the para- or ortho-position. When the group has two substituents, the meta-positions are preferred. Examples of preferred oxycarbonyl groups are methoxycarbonyl and ethoxycarbonyl.

[0019] Examples of preferred organoantimony compounds represented by the formula (1) are compounds wherein  $R^1$  and  $R^2$  are  $C_4$  alkyl,  $R^3$  and  $R^4$  are each a hydrogen atom or  $C_1$ - $C_4$  alkyl, and  $R^5$  is aryl, substituted aryl or oxycarbonyl. Especially preferable organotellurium compounds are those wherein  $R^1$  and  $R^2$  are  $C_1$ - $C_4$  alkyl,  $R^3$  and  $R^4$  are each a hydrogen atom or  $C_1$ - $C_4$  alkyl, and  $R^5$  is phenyl, substituted phenyl, methoxycarbonyl or ethoxycarbonyl.

[0020] Examples of organoantimony compounds represented by the formula (1) are as follows.

[0021] (Dimethylstibanyl-methyl)benzene, (1-dimethylstibanyl-ethyl)benzene, (2-dimethylstibanyl-propyl)benzene, 1-chloro-4-(dimethylstibanyl-methyl)benzene, 1-hydroxy-4-(dimethylstibanyl-methyl)benzene, 1-methoxy-4-(dimethylstibanyl-methyl)benzene, 1-amino-4-(dimethylstibanyl-methyl)benzene, 1-nitro-4-(dimethylstibanyl-methyl)benzene, 1-cyano-4-(dimethylstibanyl-methyl)benzene, 1-methylcarbonyl-4-(dimethylstibanyl-methyl)benzene, 1-phenylcarbonyl-4-(dimethylstibanyl-methyl)benzene, 1-methoxycarbonyl-4-(dimethylstibanyl-methyl)benzene, 1-phenoxycarbonyl-4-(dimethylstibanyl-methyl)benzene, 1-sulfonyl-4-(dimethylstibanyl-methyl)benzene, 1-trifluoromethyl-4-(dimethylstibanyl-methyl)benzene, 1-chloro-4-(1-dimethylstibanyl-ethyl)benzene, 1-hydroxy-4-(1-dimethylstibanyl-ethyl)benzene, 1-

methoxy-4-(1-dimethylstibanyl-ethyl)benzene, 1-amino-4-(1-dimethylstibanyl-ethyl)benzene, 1-nitro-4-(1-dimethylstibanyl-ethyl)benzene, 1-cyano-4-(1-dimethylstibanyl-ethyl)benzene, 1-methylcarbonyl-4-(1-dimethylstibanyl-ethyl)benzene, 1-phenylcarbonyl-4-(1-dimethylstibanyl-ethyl)benzene, 1-methoxycarbonyl-4-(1-dimethylstibanyl-ethyl)benzene, 1-phenoxy-4-(1-dimethylstibanyl-ethyl)benzene, 1-sulfonyl-4-(1-dimethylstibanyl-ethyl)benzene, 1-trifluoromethyl-4-(1-dimethylstibanyl-ethyl)benzene, 1-(1-dimethylstibanyl-ethyl)-3,5-bis-trifluoromethylbenzene, 1,2,3,4,5-pentafluoro-6-(1-dimethylstibanyl-ethyl)benzene, 1-chloro-4-(2-dimethylstibanyl-propyl)benzene, 1-hydroxy-4-(2-dimethylstibanyl-propyl)benzene, 1-methoxy-4-(2-dimethylstibanyl-propyl)benzene, 1-amino-4-(2-dimethylstibanyl-propyl)benzene, 1-nitro-4-(2-dimethylstibanyl-propyl)benzene, 1-cyano-4-(2-dimethylstibanyl-propyl)benzene, 1-methylcarbonyl-4-(2-dimethylstibanyl-propyl)benzene, 1-phenylcarbonyl-4-(2-dimethylstibanyl-propyl)benzene, 1-methoxycarbonyl-4-(2-dimethylstibanyl-propyl)benzene, 1-phenoxy-4-(2-dimethylstibanyl-propyl)benzene, 1-sulfonyl-4-(2-dimethylstibanyl-propyl)benzene, 1-trifluoromethyl-4-(2-dimethylstibanyl-propyl)benzene, 2-(dimethylstibanyl-methyl)pyridine, 2-(1-dimethylstibanyl-ethyl)pyridine, 2-(2-dimethylstibanyl-propyl)pyridine, methyl 2-dimethylstibanyl-acetate, methyl 2-dimethylstibanyl-propionate, methyl 2-dimethylstibanyl-2-methylpropionate, ethyl 2-dimethylstibanyl-acetate, ethyl 2-dimethylstibanyl propionate, ethyl 2-dimethylstibanyl-2-methylpropionate [ethyl-2-methyl-2-dimethylstibanyl-propionate], ethyl 2-(di-n-butylstibanyl)-2-methylpropionate [ethyl-2-methyl-2-di-n-butylstibanyl-propionate], 2-dimethylstibanyl-acetonitrile, 2-dimethylstibanyl-propionitrile, 2-methyl-2-dimethylstibanyl-propionitrile, (diphenylstibanyl-methyl)benzene, (1-diphenylstibanyl-ethyl)benzene, (2-diphenylstibanyl-propyl)benzene, etc. The above compounds also include all compounds having diethylstibanyl, 1-diethylstibanyl, 2-diethylstibanyl, di-n-butylstibanyl, 1-di-n-butylstibanyl, 2-di-n-butylstibanyl, diphenylstibanyl, 1-diphenylstibanyl or 2-diphenylstibanyl, as changed from the portion of dimethylstibanyl, 1-dimethylstibanyl or 2-dimethylstibanyl.

**[0022]** Preferable are (dimethylstibanyl-methyl)benzene, (1-dimethylstibanyl-ethyl)benzene, (2-dimethylstibanyl-propyl)benzene, 1-chloro-4-(dimethylstibanyl-methyl)benzene, 1-trifluoromethyl-4-(1-dimethylstibanyl-ethyl)benzene, methyl 2-dimethylstibanyl-2-methylpropionate, ethyl 2-dimethylstibanyl-2-methylpropionate [ethyl-2-methyl-2-dimethylstibanyl-propionate], ethyl 2-(di-n-butylstibanyl)-2-methylpropionate [ethyl-2-methyl-2-di-n-butylstibanyl-propionate], 1-(1-dimethylstibanyl-ethyl)-3,5-bis-trifluoromethylbenzene, 1,2,3,4,5-pentafluoro-6-(1-dimethylstibanyl-ethyl)benzene, 2-dimethylstibanyl-propionitrile, 2-methyl-2-dimethylstibanyl-propionitrile, (diethylstibanyl-methyl)benzene, (1-diethylstibanyl-ethyl)benzene, (2-diethylstibanyl-propyl)benzene, methyl 2-diethylstibanyl-2-methylpropionate, ethyl 2-diethylstibanyl-2-methylpropionate, 2-diethylstibanyl-propionitrile, 2-methyl-2-diethylstibanyl-propionitrile, (di-n-butylstibanyl-methyl)benzene, (1-di-n-butylstibanyl-ethyl)benzene, (2-di-n-butylstibanyl-propyl)benzene, methyl 2-di-n-butylstibanyl-2-methylpropionate, ethyl 2-di-n-butylstibanyl-2-methylpropionate, 2-di-n-butylstibanyl propionitrile, 2-methyl-2-di-n-butylstibanyl propionitrile.

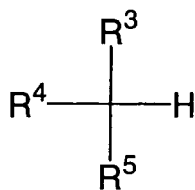
**[0023]** The organoantimony compound represented by the formula (1) can be prepared by reacting a compound of the formula (3) and, a compound of the formula (4) or a compound of the formula (5)



(wherein  $\text{R}^1$  and  $\text{R}^2$  are defined above, and Z is a halogen atom or alkali metal)



(wherein  $\text{R}^3$ ,  $\text{R}^4$  and  $\text{R}^5$  are defined above, and X is a halogen atom)



(5)

10 (wherein  $R^3$ ,  $R^9$  and  $R^5$  are defined above).

[0024] Examples of compounds represented by the formula (3) are as follows

[0025] Examples of groups represented by  $R^1$  and  $R^2$  are as given above.

[0026] Examples of groups represented by Z are halogen atom such as fluorine, chlorine, bromine or iodine or alkali metal such as sodium, potassium or lithium. Preferable are chlorine, bromine, sodium and lithium.

15 [0027] Examples of compound (3) are dimethylstibanyl bromide, diethylstibanyl bromide, di-n-butylstibanyl bromide, diphenylstibanyl bromide, dimethylstibanyl sodium, diethylstibanyl sodium, di-n-butylstibanyl sodium and diphenylstibanyl sodium. The above compounds also include all compounds having chloride or iodide, as changed from the portion of bromide. Further, the above compounds also include all compounds having potassium or lithium, as changed from the portion of sodium.

20 [0028] Examples of compounds represented by the formula (4) are as follows

[0029] Examples of groups represented by  $R^3$ ,  $R^4$  and  $R^5$  are as given above.

[0030] Examples of groups represented by X can be a halogen atom such as fluorine, chlorine, bromine or iodine. Chlorine and bromine are preferable.

25 [0031] Examples of compounds usable are benzyl chloride, benzyl bromide, 1-chloro-1-phenylethane, 1-bromo-1-phenylethane, 2-chloro-2-phenylpropane, 2-bromo-2-phenylpropane, p-chlorobenzyl chloride, p-hydroxybenzyl chloride, p-methoxybenzyl chloride, p-aminobenzyl chloride, p-nitrobenzyl chloride, p-cyanobenzyl chloride, p-methylcarbonylbenzyl chloride, phenylcarbonylbenzyl chloride, p-methoxycarbonylbenzyl chloride, p-phenoxy carbonylbenzyl chloride, p-sulfonylbenzyl chloride, p-trifluoromethylbenzyl chloride, 1-chloro-1-(p-chlorophenyl)ethane, 1-bromo-1-(p-chlorophenyl)ethane, 1-chloro-1-(p-hydroxyphenyl)ethane, 1-bromo-1-(p-hydroxyphenyl)-ethane, 1-chloro-1-(p-methoxyphenyl)ethane, 1-bromo-1-(p-methoxyphenyl)ethane, 1-chloro-1-(p-aminophenyl)ethane, 1-bromo-1-(p-aminophenyl)ethane, 1-chloro-1-(p-nitrophenyl)ethane, 1-bromo-1-(p-nitrophenyl)ethane, 1-chloro-1-(p-cyanophenyl)ethane, 1-bromo-1-(p-cyanophenyl)ethane, 1-chloro-1-(p-methylcarbonylphenyl)ethane, 1-bromo-1-(p-methylcarbonylphenyl)ethane, 1-chloro-1-(p-phenylcarbonylphenyl)ethane, 1-bromo-1-(p-phenylcarbonylphenyl)-ethane, 1-chloro-1-(p-methoxycarbonylphenyl)ethane, 1-bromo-1-(p-methoxycarbonylphenyl)ethane, 1-chloro-1-(p-phenoxy carbonylphenyl)-ethane, 1-bromo-1-(p-phenoxy carbonylphenyl)ethane, 1-chloro-1-(p-sulfonylphenyl)ethane, 1-bromo-1-(p-sulfonylphenyl)ethane, 1-chloro-1-(p-trifluoromethylphenyl)ethane, 1-bromo-1-(p-trifluoromethylphenyl)ethane, 2-chloro-2-(p-chlorophenyl)propane, 2-bromo-2-(p-chlorophenyl)propane, 2-chloro-2-(p-hydroxyphenyl)-propane, 2-bromo-2-(p-hydroxyphenyl)propane, 2-chloro-2-(p-methoxyphenyl)propane, 2-bromo-2-(p-methoxyphenyl)propane, 2-chloro-2-(p-aminophenyl)propane, 2-bromo-2-(p-aminophenyl)propane, 2-chloro-2-(p-nitrophenyl)propane, 2-bromo-2-(p-nitrophenyl)-propane, 2-chloro-2-(p-cyanophenyl)propane, 2-bromo-2-(p-cyanophenyl)propane, 2-chloro-2-(p-methylcarbonylphenyl)propane, 2-bromo-2-(p-methylcarbonylphenyl)propane, 2-chloro-2-(p-phenylcarbonylphenyl)propane, 2-bromo-2-(p-phenylcarbonylphenyl)-propane, 2-chloro-2-(p-methoxycarbonylphenyl)propane, 2-bromo-2-(p-methoxycarbonylphenyl)propane, 2-chloro-1-(p-phenoxy carbonylphenyl)propane, 2-bromo-2-(p-phenoxy carbonylphenyl)propane, 2-chloro-2-(p-sulfonylphenyl)propane, 2-bromo-2-(p-sulfonylphenyl)propane, 2-chloro-2-(p-trifluoromethylphenyl)propane, 2-bromo-2-(p-trifluoromethylphenyl)propane, 2-(chloromethyl)pyridine, 2-(bromomethyl)pyridine, 2-(1-chloroethyl)pyridine, 2-(1-bromoethyl)pyridine, 2-(2-chloropropyl)pyridine, 2-(2-bromopropyl)pyridine, methyl 2-chloroacetate, methyl 2-bromoacetate, methyl 2-chloropropionate, methyl 2-bromoacetate, methyl 2-chloro-2-methylpropionate, methyl 2-bromo-2-methylpropionate, ethyl 2-chloroacetate, ethyl 2-bromoacetate, ethyl 2-chloropropionate, ethyl 2-bromoethanoate, ethyl 2-chloro-2-ethylpropionate, ethyl 2-bromo-2-ethylpropionate, 2-chloroacetonitrile, 2-bromoacetonitrile, 2-chloropropionitrile, 2-bromopropionitrile, 2-chloro-2-methylpropionitrile, 2-bromo-2-methylpropionitrile, (1-bromoethyl)benzene, ethyl-2-bromo-isobutyrate, 1-(1-bromoethyl)-4-chlorobenzene, 1-(1-bromoethyl)-4-trifluoromethylbenzene, 1-(1-bromoethyl)-3,5-bis-trifluoromethylbenzene, 1,2,3,4,5-pentafluoro-6-(1-bromoethyl)benzene, 1-(1-bromoethyl)-4-methoxybenzene, ethyl-2-bromo-isobutyrate, etc.

[0032] Examples of compounds represented by the formula (5) are as follows

55 [0033] Examples of groups represented by  $R^3$ ,  $R^4$  and  $R^5$  are as given above.

[0034] Examples of compounds usable are those wherein halogen atom is replaced by hydrogen atom in the compound represented by the formula (4).

[0035] Next, a detailed description will be given of the process for preparing the compound (1).

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(A) Process of using the compound (3) and the compound (5):

[0036] The compound (5) is dissolved in a solvent. Examples of solvents usable are N,N-dimethylformamide (DMF), diakyl ether, tetrahydrofuran (THF), dimethoxyethane and like ethers, toluene, xylene and like aromatic solvents, hexane and like aliphatic hydrocarbons. THF is preferable. The amount of solvent to be used, which is suitably adjusted, is 1 to 100 ml, preferably 5 to 20 ml, per gram of the compound (5).

[0037] To the solution is slowly added dropwise lithium diisopropylamide (LDA), lithium hexamethyldisilazide, lithium 2,2,6,6-tetramethylpiperizide or like lithium amide compound, potassium diisopropylamide, potassium hexamethyldisilazide, potassium 2,2,6,6-tetramethylpiperizide, potassium amide (KNH<sub>2</sub>) or like potassium amide compound, or sodium amide, followed by stirring. The reaction time differs with the reaction temperature and pressure and is usually 5 minutes to 24 hours, preferably 10 minutes to 2 hours. The reaction temperature is -150°C to 80°C, preferably -100°C to 80°C, more preferably -78°C to 80°C, most preferably -78°C to 20°C. The reaction is carried out usually under atmospheric pressure, but may be carried out at increased pressure or in a vacuum. Next, a compound (3) is added to the reaction mixture, followed by stirring. The reaction time differs with the reaction temperature and pressure and is usually 5 minutes to 24 hours, preferably 10 minutes to 2 hours. The reaction temperature is -78°C to 80°C, preferably -78°C to 20°C, more preferably -50°C to 20°C. The reaction is carried out usually under atmospheric pressure, but may be carried out at increased pressure or in a vacuum. The order of the compounds to be added may be changed suitably depending on the desired compound.

[0038] The proportion of the compound (5) to the compound (3) is 0.5 to 1.5 moles, preferably 0.8 to 1.2 moles of the compound (5), per mole of the compound (3).

[0039] After the completion of the reaction, the solvent is concentrated, and the desired compound is isolated and purified. Although the method of purification can be determined suitably depending on the compound, usually vacuum distillation or recrystallization is preferable.

(B) Process of using the compound (3) and the compound (4):

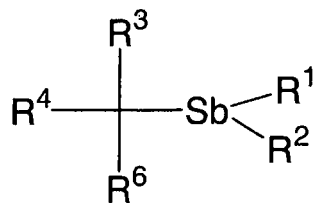
[0040] The compound (3) is dissolved in a solvent. Examples of solvents usable are liquid ammonia, mixed solvent of liquid ammonia and tetrahydrofuran, mixed solvent of liquid ammonia and ether, mixed solvent of liquid ammonia and 1,4-dioxane or the like. The amount of solvent to be used, which is suitably adjusted, is 1 to 100 ml, preferably 5 to 20 ml, per gram of the compound (3).

[0041] To the solution is slowly added dropwise metal magnesium, metal sodium, metal potassium, metal lithium, sodium bromide, ammonium bromide or the like, followed by stirring. The reaction temperature is -78°C to 30°C, preferably -78°C to 0°C. The reaction is carried out usually under atmospheric pressure, but may be carried out at increased pressure or in a vacuum. The order of the compounds to be added may be changed suitably depending on the desired compound.

[0042] Next, a compound (4) is added to the reaction mixture, followed by stirring. The reaction time differs with the reaction temperature and pressure and is usually 5 minutes to 24 hours, preferably 10 minutes to 2 hours. The reaction temperature is -78°C to 30°C, preferably -78°C to 0°C. The reaction is carried out usually under atmospheric pressure, but may be carried out at increased pressure or in a vacuum.

[0043] The proportion of the compound (4) to the compound (3) is 0.5 to 1.5 moles, preferably 0.8 to 1.2 moles of the compound (4), per mole of the compound (3). After the completion of the reaction, the solvent is concentrated, and the desired compound is isolated and purified. Although the method of purification can be determined suitably depending on the compound, usually vacuum distillation or recrystallization is preferable.

[0044] A living radical polymerization initiator of the present invention is represented by the formula (2)



(2)

wherein R<sup>1</sup> and R<sup>2</sup> are C<sub>1</sub>-C<sub>8</sub> alkyl, aryl, substituted aryl or an aromatic heterocyclic group, R<sup>3</sup> and R<sup>4</sup> are each a hydrogen atom or C<sub>1</sub>-C<sub>8</sub> alkyl, and R<sup>5</sup> is aryl, substituted aryl, an aromatic heterocyclic group, acyl, oxycarbonyl or cyano.

[0045] Examples of groups represented by R<sup>1</sup> to R<sup>4</sup> are as given above. Examples of groups represented by R<sup>6</sup> are those given in R<sup>5</sup> and acyl group.

**[0046]** Examples of acyl groups usable are formyl, acetyl and benzoyl.

**[0047]** An azo type polymerization initiator used in the present invention is not particularly limited insofar as it is usable in a usual radical polymerization. Example thereof are 2,2'-azobis-isobutyronitrile (AIBN), 2,2'-azobis(2-methylbutyronitrile) (AMBN), 2,2'-azobis(2,4-dimethylvaleronitrile) (ADV N), 1,1'-azobis(1-cyclohexanecarbonitrile) (ACHN), dimethyl-2,2'-azobisisobutyrate (MAIB), 4,4'-azobis(4-cyanovaleric acid) (ACVA), 1,1'-azobis(1-acetoxy-1-phenylethane), 2,2'-azobis(2-methylbutylamide), 2,2'-azobis(4-methoxy-2,4-dimethylvaleronitrile), 2,2'-azobis(2-methylamidinopropane) dihydrochloride, 2,2'-azobis[2-(2-imidazoline-2-yl)propane], 2,2'-azobis[2-methyl-N-(2-hydroxyethyl)propionamide], 2,2'-azobis(2,4,4-trimethylpentane), 2-cyano-2-propylazofornamide, 2,2'-azobis(N-butyl-2-methylpropionamide) and 2,2'-azobis(N-cyclohexyl-2-methylpropionamide).

**[0048]** These azo type polymerization initiators are preferably selected depending on the reaction conditions. For example, in case of low temperature polymerization, preferable are 2,2'-azobis(2,4-dimethylvaleronitrile) (ADV N) and 2,2'-azobis(4-methoxy-2,4-dimethylvaleronitrile). In case of middle temperature polymerization, preferable are 2,2'-azobis-isobutyronitrile (AIBN), 2,2'-azobis(2-methylbutyronitrile) (AMBN), dimethyl-2,2'-azobisisobutyrate (MAIB) and 1,1'-azobis(1-acetoxy-1-phenylethane). In case of high temperature polymerization, preferable are 1,1'-azobis(1-cyclohexanecarbonitrile) (ACHN), 2-cyano-2-propylazofornamide, 2,2'-azobis(N-butyl-2-methylpropionamide), 2,2'-azobis(N-cyclohexyl-2-methylpropionamide) and 2,2'-azobis(2,4,4-trimethylpentane). In case of using aqueous solvent, preferable are 4,4'-azobis(4-cyanovaleric acid) (ACVA), 2,2'-azobis(2-methylbutylamide), 2,2'-azobis(2-methylamidinopropane) dihydrochloride, 2,2'-azobis[2-(2-imidazoline-2-yl)propane] and 2,2'-azobis[2-methyl-N-(2-hydroxyethyl)propionamide].

**[0049]** The vinyl monomer to be used in the present invention is not particularly limited insofar as the monomer can be subjected to radical polymerization. Examples of vinyl monomers usable are as follows.

**[0050]** Methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, butyl (meth)acrylate, octyl (meth)acrylate, lauryl (meth)acrylate, (meth)acrylic acid 2-hydroxyethyl ester and like (meth)acrylic acid esters, cyclohexyl (meth)acrylate, methylcyclohexyl (meth)acrylate, isobornyl (meth)acrylate, cyclododecyl (meth)acrylate and like cycloalkyl-containing unsaturated monomers.

**[0051]** (Meth)acrylic acid, maleic acid, fumaric acid, itaconic acid, citraconic acid, crontonic acid, maleic anhydride, methyl and like carboxyl-containing unsaturated monomers.

**[0052]** N,N-Dimethylaminopropyl(meth)acrylamide, N,N-dimethylaminoethyl(meth)acrylamide, 2-(dimethylamino)ethyl (meth)acrylate, N,N-dimethylaminopropyl (meth)acrylate and like unsaturated monomers containing a tertiary amine.

**[0053]** N-2-Hydroxy-3-acryloyloxypropyl-N,N,N-trimethylammonium chloride, N-methacryloylaminoethyl-N,N,N-dimethylbenzylammonium chloride and like unsaturated monomers containing quaternary ammonium base.

**[0054]** Glycidyl (meth)acrylate and like epoxy-containing unsaturated monomers.

**[0055]** Styrene,  $\alpha$ -methylstyrene, 4-methylstyrene (p-methylstyrene), 2-methylstyrene (o-methylstyrene), 3-methylstyrene (m-methylstyrene), 4-methoxystyrene (p-methoxystyrene), p-t-butylstyrene, p-n-butylstyrene, p-tert-butoxystyrene, 2-hydroxymethylstyrene, 2-chlorostyrene (o-chlorostyrene), 4-chlorostyrene (p-chlorostyrene), 2,4-dichlorostyrene, 1-vinylnaphthalene, divinylbenzene, p-styrenesulfonic acid or an alkali metal salt thereof (sodium salt or potassium salt, etc.) and like aromatic unsaturated monomers (styrene type monomer).

**[0056]** 2-Vinylthiophene, N-methyl-2-vinylpyrrole, 1-vinyl-2-pyrrolidone, 2-vinylpyridine, 4-vinylpyridine and like unsaturated monomers containing a heterocyclic ring.

**[0057]** N-Vinylformaldehyde, N-vinylacetamide and like vinylamides.

**[0058]** (Meth)acrylamide, N-methyl(meth)acrylamide, N-isopropyl(meth)acrylamide, N,N-dimethyl(meth)acrylamide, sodium 2-acrylamide-2-methylpropanesulfonate, 6-acrylamide hexanoic acid and like (meth)acrylamide type monomers.

**[0059]** 1-Hexene, 1-octene, 1-decene and like  $\alpha$ -olefins.

**[0060]** Butadiene, isoprene, 4-methyl-1,4-hexadiene, 7-methyl-1,6-octadiene and like dienes.

**[0061]** Vinyl acetate, vinyl benzoate and like vinyl carboxylate.

**[0062]** Hydroxyethyl (meth)acrylate, (meth)acrylonitrile, methyl vinyl ketone, vinyl chloride, vinylidene chloride.

**[0063]** Preferable among these are (meth)acrylic acid ester, unsaturated monomers containing a cycloalkyl group, aromatic unsaturated monomers (styrene type monomers), (meth)acrylamide type monomers, (meth)acrylonitrile and methyl vinyl ketone.

**[0064]** Examples of preferable (meth)acrylic acid ester monomers are methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, butyl (meth)acrylate and (meth)acrylic acid 2-hydroxyethyl ester. Especially preferable are methyl methacrylate, ethyl methacrylate, propyl methacrylate, butyl methacrylate and methacrylic acid 2-hydroxyethyl ester.

**[0065]** Examples of preferable unsaturated monomers containing a cycloalkyl group are cyclohexyl (meth)acrylate and isobornyl (meth)acrylate. Especially preferable are cyclohexyl methacrylate and isobornyl methacrylate.

**[0066]** Examples of preferable styrene type monomers are styrene,  $\alpha$ -methylstyrene, o-methylstyrene, p-methylstyrene, p-methoxystyrene, p-t-butylstyrene, p-n-butylstyrene, p-tert-butylstyrene, p-chlorostyrene, and p-styrenesulfonic acid or an alkali metal salt thereof (sodium salt or potassium salt, etc.). More preferable are styrene and p-chlorostyrene.

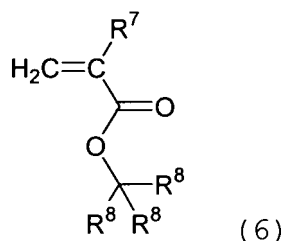
**[0067]** Example of preferable (meth)acrylamide type monomers is N-isopropyl(meth)acrylamide. Especially preferable



is N-isopropyl-methacrylamide.

**[0068]** The term "(meth)acrylic acid" refers collectively to "acrylic acid" and "methacrylic acid."

**[0069]** Also useful are vinyl monomers of the formula (6)



wherein  $\text{R}^7$  is hydrogen, methyl, trifluoromethyl or hydroxymethyl, and the groups  $\text{R}^8$  are each independently of the other a monovalent alicyclic hydrocarbon group having 4 to 20 carbon atoms or a derivative thereof, or straight-chain or branched alkyl having 1 to 4 carbon atoms, at least one of the groups  $\text{R}^8$  is the alicyclic hydrocarbon group or a derivative thereof, or two of the groups  $\text{R}^8$ , when taken together with the carbon atom to which they are attached, form a bivalent alicyclic hydrocarbon group having 4 to 20 carbon atoms or a derivative thereof, and the remaining group or groups  $\text{R}^8$  being straight-chain or branched alkyl having 1 to 4 carbon atoms or a monovalent alicyclic hydrocarbon group having 4 to 20 carbon atoms or a derivative thereof.

**[0070]** Examples of monovalent alicyclic hydrocarbon groups having 4 to 20 carbon atoms or derivatives thereof represented by  $\text{R}^8$ , examples of alicyclic hydrocarbon groups or derivatives thereof represented by at least one of the groups  $\text{R}^8$ , or examples of bivalent alicyclic hydrocarbon groups having 4 to 20 carbon atoms or derivatives thereof and represented by two of the groups  $\text{R}^8$  as taken together with the carbon atom to which they are attached are groups comprising an alicyclic ring and derived from cycloalkanes such as bicyclo[2.2.1]heptane, tricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl, tetracyclo[6.2.1.3<sup>6,0</sup>.0<sup>2,7</sup>]dodeca-4-yl, adamantane, cyclopentane and cyclohexane; and groups comprising such an alicyclic ring and substituted with at least one kind of or at least one of straight-chain, branched or cyclic alkyl groups having 1 to 4 carbon atoms, such as methyl, ethyl, n-propyl, i-propyl, n-butyl, 2-methylpropyl, 1-methylpropyl and tert-butyl.

**[0071]** Examples of derivatives of monovalent or bivalent alicyclic hydrocarbon groups represented by  $\text{R}^8$  are groups having at least one kind of or at least one of substituents including hydroxyl; carboxyl; oxo (i.e., the group =O); hydroxyalkyl groups having 1 to 4 carbon atoms, such as hydroxymethyl, 1-hydroxyethyl, 2-hydroxyethyl, 1-hydroxypropyl, 2-hydroxypropyl, 3-hydroxypropyl, 2-hydroxybutyl, 3-hydroxybutyl and 4-hydroxybutyl; alkoxy groups having 1 to 4 carbon atoms, such as methoxy, ethoxy, n-propoxy, i-propoxy, n-butoxy, 2-methylpropoxy, 1-methylpropoxy and tert-butoxy; cyano; and cyanoalkyl groups having 2 to 4 carbon atoms, such as cyanomethyl, 2-cyanomethyl, 3-cyanopropyl and 4-cyanobutyl. Preferable among these substituents are hydroxyl, carboxyl, hydroxymethyl, cyano, cyanomethyl, etc.

**[0072]** Examples of straight-chain or branched alkyl groups having 1 to 4 carbon atoms and represented by  $\text{R}^8$  are methyl, ethyl, n-propyl, i-propyl, n-butyl, 2-methylpropyl, 1-methylpropyl, tert-butyl, etc. Preferable among these alkyl groups are methyl, ethyl, n-propyl and i-propyl.

**[0073]** Given below are examples of preferred functional group side chains forming  $-\text{C}(\text{R}^8)_3$  in the formula (6).

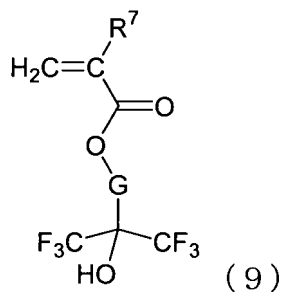
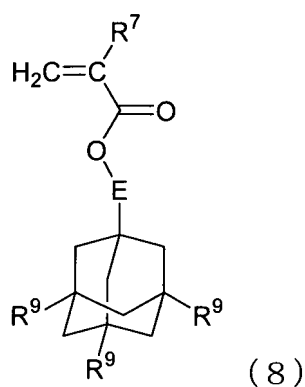
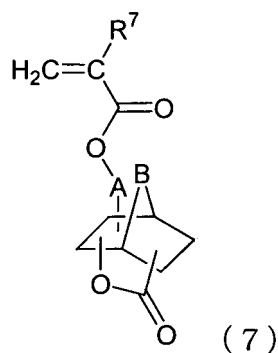
**[0074]** 1-Methyl-1-cyclopentyl, 1-ethyl-1-cyclopentyl, 1-methyl-1-cyclohexyl, 1-ethyl-1-cyclohexyl, 2-methyladamantane-2-yl, 2-methyl-3-hydroxyadamantane-2-yl, 2-ethyladamantane-2-yl, 2-ethyl-3-hydroxyadamantane-2-yl, 2-n-propyladamantane-2-yl, 2-n-propyl-3-hydroxyadamantane-2-yl, 2-isopropyladamantane-2-yl, 2-isopropyl-3-hydroxyadamantane-2-yl, 2-methylbicyclo[2.2.1]hept-2-yl, 2-ethylbicyclo[2.2.1]hept-2-yl, 8-methyltricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl, 8-ethyltricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl, 4-methyltetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl, 4-ethyltetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl, 1-(bicyclo[2.2.1]hept-2-yl)-1-methylethyl, 1-(tricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl)-1-methylethyl, 1-(tetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl)-1-methylethyl, 1-(adamantane-1-yl)-1-methylethyl, 1-(3-hydroxyadamantane-1-yl)-1-methylethyl, 1,1-dicyclohexylethyl, 1,1-di(bicyclo[2.2.1]hept-2-yl)ethyl, 1,1-di(tricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl)ethyl, 1,1-di(tetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl)ethyl, 1,1-di(adamantane-1-yl)ethyl, etc.

**[0075]** Further, preferable examples of the vinyl monomers of the formula (6) are as follows. (Meth)acrylic acid 1-methyl-1-cyclopentyl ester, (meth)acrylic acid 1-ethyl-1-cyclopentyl ester, (meth)acrylic acid 1-methyl-1-cyclohexyl ester, (meth)acrylic acid 1-ethyl-1-cyclohexyl ester, (meth)acrylic acid 2-methyladamantane-2-yl ester, (meth)acrylic acid 2-methyl-3-hydroxyadamantane-2-yl ester, (meth)acrylic acid 2-ethyladamantane-2-yl ester, (meth)acrylic acid 2-ethyl-3-hydroxyadamantane-2-yl ester, (meth)acrylic acid 2-n-propyladamantane-2-yl ester, (meth)acrylic acid 2-n-propyl-3-hydroxyadamantane-2-yl ester, (meth)acrylic acid 2-isopropyladamantane-2-yl ester, (meth)acrylic acid 2-isopropyl-3-hydroxyadamantane-2-yl ester, (meth)acrylic acid 2-methyladamantane-2-yl ester, (meth)acrylic acid 2-methylbicyclo[2.2.1]hept-2-yl ester, (meth)acrylic acid 2-ethylbicyclo[2.2.1]hept-2-yl ester, (meth)acrylic acid 8-methyltricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl ester, (meth)acrylic acid 8-ethyltricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl ester, (meth)acrylic acid 4-methyltetra-

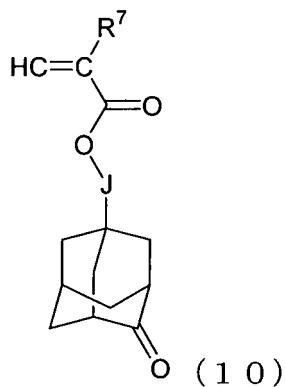
cyclo [6.2.1.1<sup>2,6</sup>.0<sup>2,7</sup>]dodeca-4-yl ester, (meth)acrylic acid 4-ethyltetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl ester, (meth) acrylic acid 1-(bicyclo[2.2.1]hept-2-yl)-1-methyl ester, (meth)acrylic acid 1-(tricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl)-1-methyl ester, (meth)acrylic acid 1-(tetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl)-1-methylethyl ester, (meth)acrylic acid 1-(adamantane-1-yl)-1-methylethyl ester, (meth)acrylic acid 1-(3-hydroxyadamantane-1-yl)-1-methylethyl ester, (meth)acrylic acid 1,1-dicyclohexylethyl ester, (meth)acrylic acid 1,1-di(bicyclo[2.2.1]hept-2-yl)ethyl ester, (meth)acrylic acid 1,1-di(tricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl)ethyl ester, (meth)acrylic acid 1,1-di(tetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl)ethyl ester and (meth) acrylic acid 1,1-di(adamantane-1-yl)ethyl ester.

**[0076]** Especially preferable examples of the vinyl monomers of the formula (6) are (meth)acrylic acid 1-methyl-1-cyclopentyl ester, (meth)acrylic acid 1-ethyl-1-cyclopentyl ester, (meth)acrylic acid 1-methyl-1-cyclohexyl ester, (meth) acrylic acid 1-ethyl-1-cyclohexyl ester, (meth)acrylic acid 2-methyladamantane-2-yl ester, (meth)acrylic acid 2-ethyladamantane-2-yl ester, (meth)acrylic acid 2-n-propyladamantane-2-yl ester, (meth)acrylic acid 2-isopropyladamantane-2-yl ester and (meth)acrylic acid 1-(adamantane-1-yl)-1-methylethyl ester.

**[0077]** Vinyl monomers of the formulas (7) to (13) below are usable in addition to the vinyl monomers of the formula (6)



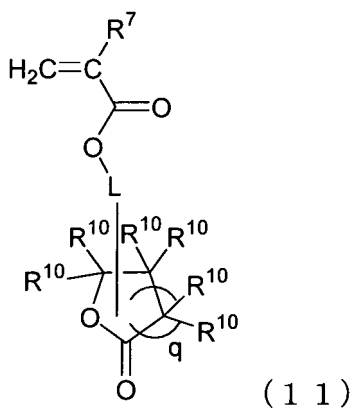
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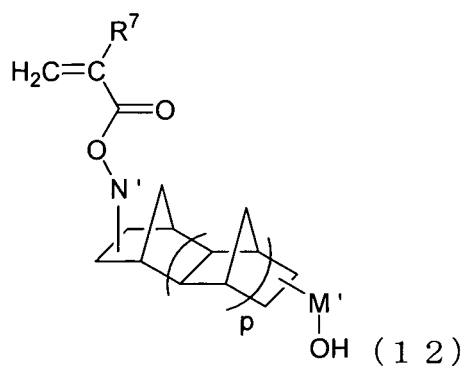
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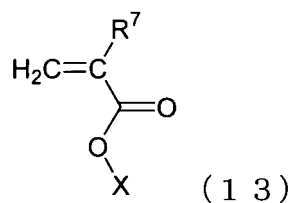
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[0078] In the above formulae (7) to (13), R<sup>7</sup> is the same as the R<sup>7</sup> of the formula (6).

[0079] In the formula (7), A is a single bond, or straight-chain or branched alkylene which may have a substituent with 1 to 6 carbon atoms, or a mono- or di-alkylene glycol group or alkylene ester group. Examples of straight-chain or branched alkylene groups having 1 to 6 carbon atoms are methylene, ethylene, propylene, butylene, pentylene, hexylene and cyclohexylene. B is a single bond or alkylene which may have a substituent with 1 to 3 carbon atoms, alkyloxy or

an oxygen atom.

**[0080]** In the formula (8), E is a single bond or bivalent alkylene having 1 to 3 carbon atoms, and the groups R<sup>9</sup> are independent of one another and are each hydroxyl, cyano, carboxyl, -COOR<sup>c</sup> or -Y-R<sup>d</sup>. R<sup>c</sup> is a hydrogen atom, straight-chain or branched alkyl having 1 to 4 carbon atoms or alicyclic alkyl having 3 to 20 carbon atoms. The groups Y are independent of one another and are each a single bond or bivalent alkylene having 1 to 3 carbon atoms. The groups R<sup>4</sup> are independent of one another and are each a hydrogen atom, hydroxyl, cyano or the group -COOR<sup>e</sup>. However, at least one of the groups R<sup>9</sup> is not a hydrogen atom. Examples of E and Y are a single bond, methylene, ethylene and propylene.

**[0081]** The group R<sup>e</sup> in -COOR<sup>e</sup> is a hydrogen atom, straight-chain or branched alkyl having 1 to 4 carbon atoms or alicyclic alkyl having 3 to 20 carbon atoms. Examples of straight-chain or branched alkyl groups having 1 to 4 carbon atoms can be methyl, ethyl, n-propyl, i-propyl, n-butyl, 2-methylpropyl, 1-methylpropyl and tert-butyl. Examples of alicyclic alkyl having 3 to 20 carbon atoms are cycloalkyl groups represented by -C<sub>n</sub>H<sub>2n-1</sub> (wherein n is an integer of 3 to 20), such as cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl and cyclooctyl; polycyclic alicyclic alkyl groups such as bicyclo[2.2.1]heptyl, tricyclo[5.2.1.0<sup>2-6</sup>]decyl, tetracyclo[6.2.1<sup>3.6</sup>.0<sup>2.7</sup>]dodecanyl and adamantyl; or cycloalkyl groups or polycyclic alicyclic alkyl groups as partly substituted with at least one kind of or at least one of straight-chain, branched or cyclic alkyl groups.

**[0082]** In the formula (9), G is a single bond, straight-chain or branched alkylene having 1 to 6 carbon atoms, an alicyclic hydrocarbon group having 4 to 20 carbon atoms, alkylene glycol group or alkylene ester group. Examples of straight-chain or branched alkylene groups having 1 to 6 carbon atoms are methylene, ethylene, propylene, butylenes, pentylene, hexylene and cyclohexylene.

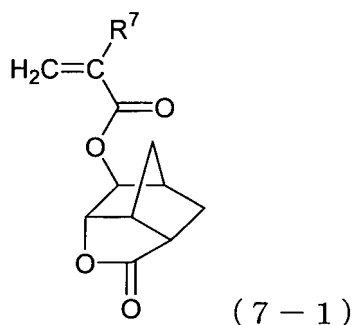
**[0083]** In the formula (10), J is a single bond, or straight-chain, branched or cyclic alkylene which may have a substituent with 1 to 20 carbon atoms, an alkylene glycol group or alkylene ester group.

**[0084]** In the formula (11), L is a single bond, or straight-chain, branched or cyclic alkylene which may have a substituent with 1 to 20 carbon atoms, an alkylene glycol group or alkylene ester group. R<sup>10</sup> is a hydrogen atom, straight-chain or branched alkyl having 1 to 4 carbon atoms, alkoxy, hydroxyalkyl, a bivalent alicyclic hydrocarbon group having 3 to 20 carbon atoms or a group comprising a derivative thereof. q is 1 or 2.

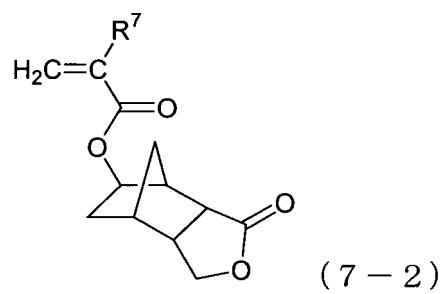
**[0085]** In the formula (12), N' and M' are independent of each other and are each straight-chain, branched or cyclic alkylene which may have a substituent with 1 to 20 carbon atoms, an alkylene glycol group or alkylene ester group. Examples of straight-chain or branched alkylene groups are methylene, ethylene, propylene, butylenes, pentylene, hexylene and cyclohexylene. p is 0 or 1.

**[0086]** In the formula (13), X is a polycyclic alicyclic hydrocarbon group having 7 to 20 carbon atoms, containing no polar group and consisting only of carbon and hydrogen. Examples of such polycyclic alicyclic hydrocarbon groups are hydrocarbon groups comprising an alicyclic ring and derived from cycloalkanes such as bicyclo[2.2.1]heptane, bicyclo[2.2.2]octane, tricyclo[5.2.1.0<sup>2.6</sup>]decane, tetracyclo[4.4.0.1<sup>2.6</sup>.1<sup>7.10</sup>]dodecane and tetracyclo[6.2.1.1<sup>3.6</sup>.0<sup>2.7</sup>]decane. Such alicyclic rings may have a substituent, that is, they may be substituted with at least one kind of or at least one of straight-chain or cyclic alkyl groups having 1 to 4 carbon atoms, such as methyl, ethyl, n-propyl, i-propyl, n-butyl, 2-methylpropyl, 1-methylpropyl and tert-butyl.

**[0087]** Examples of vinyl monomers of the formula (7) are those represented by the formulae (7-1) to (7-7) given below.

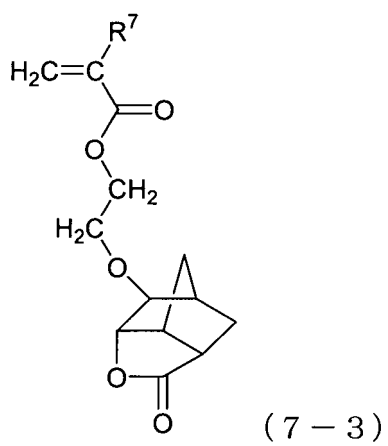


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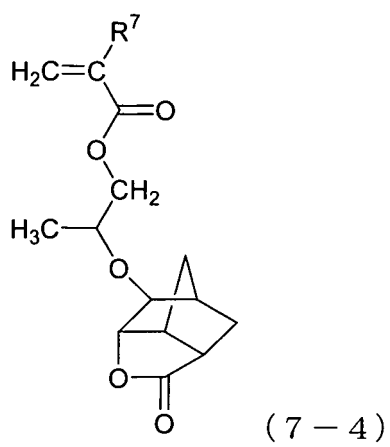
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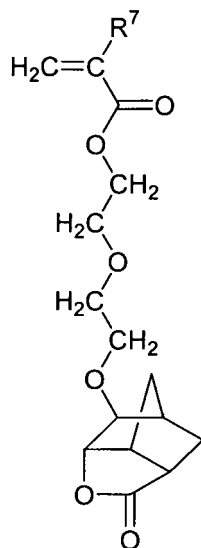
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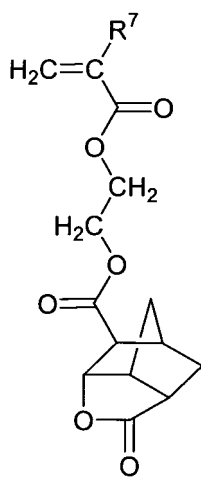
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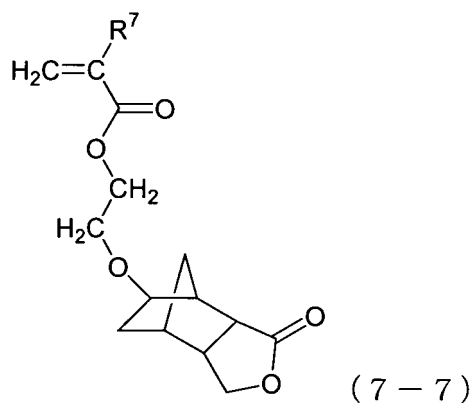
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**[0088]** R<sup>7</sup> is a hydrogen atom, methyl, trifluoromethyl or hydroxymethyl. Preferable as the vinyl monomer is norbornane wherein the methylene group at the 7-position corresponding to the bridgehead is replaced by an oxygen atom.

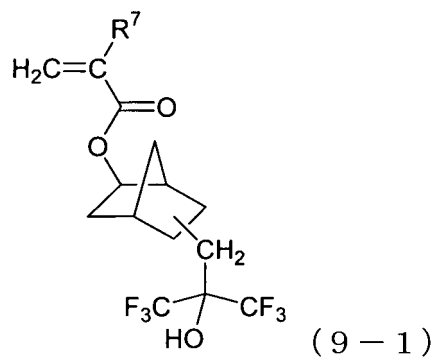
**[0089]** Given below are examples of preferred vinyl monomers of the formula (8).

**[0090]** (Meth)acrylic acid 3-hydroxyadamantane ester, (meth)acrylic acid 3-hydroxyadamantane-1-yl methyl ester, (meth)acrylic acid 3,5-dihydroxyadamantane-1-yl methyl ester, (meth)acrylic acid 3-hydroxy-5-cyanoadamantane-1-yl



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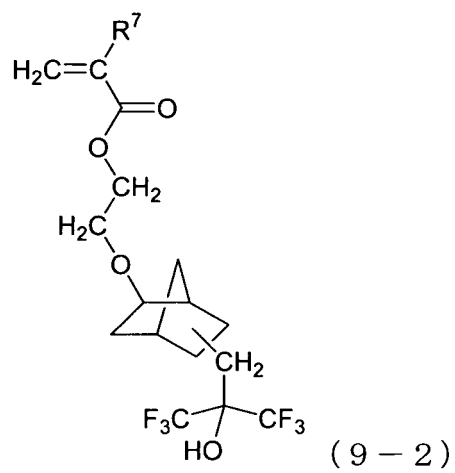
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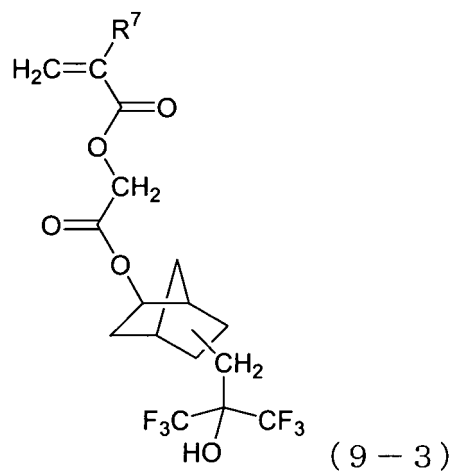


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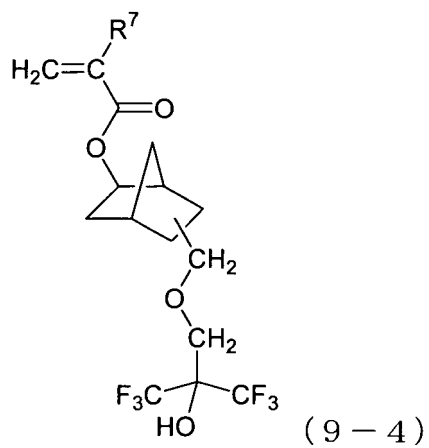
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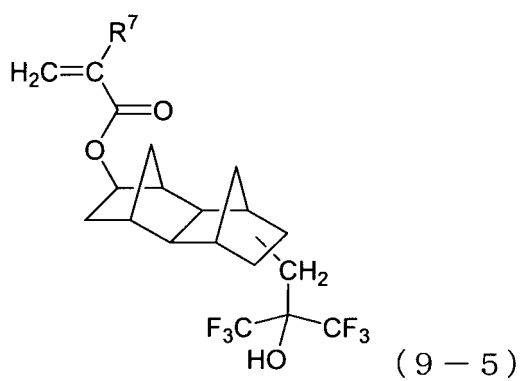
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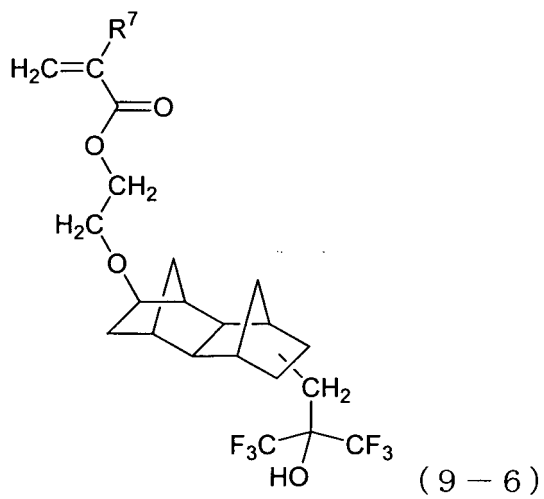
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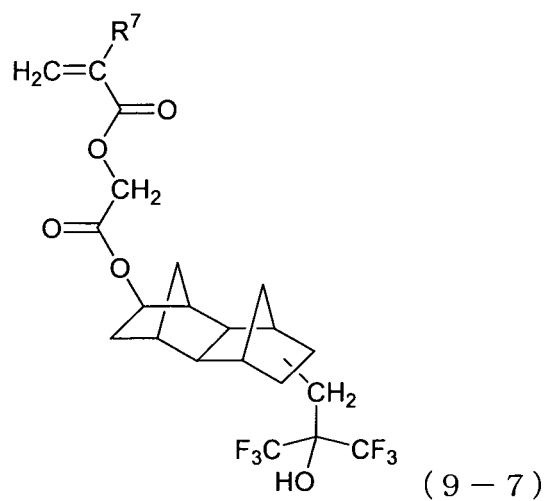
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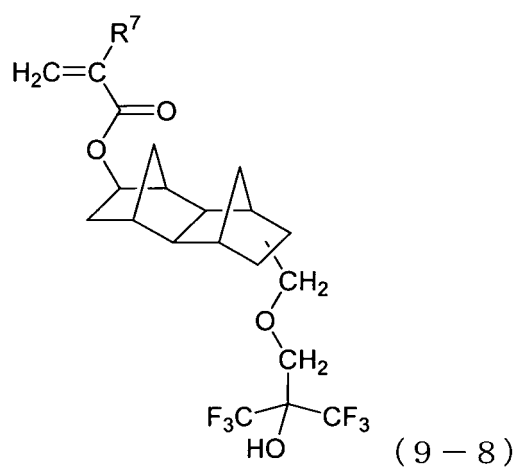
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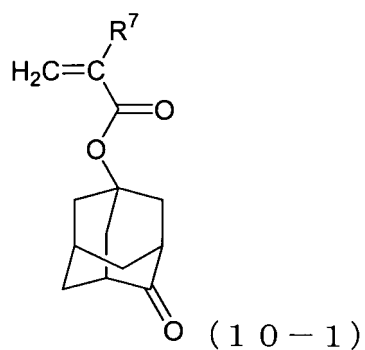
**[0093]** In the above formulae, R<sup>7</sup> is a hydrogen atom, methyl, trifluoromethyl or hydroxymethyl. Preferable as the vinyl monomer is norbornane wherein the methylene group at the 7-position corresponding to the bridgehead is replaced by an oxygen atom.

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**[0094]** Among the vinyl monomers of the formula (10), especially preferable are those of the formulas (10-1) to (10-4) below.

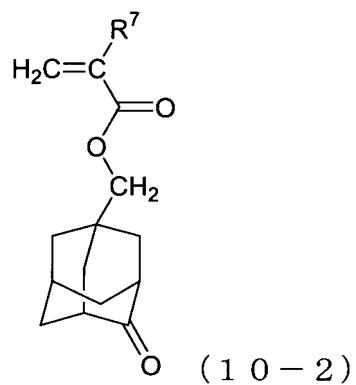
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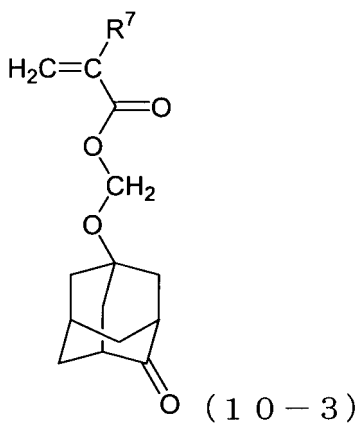
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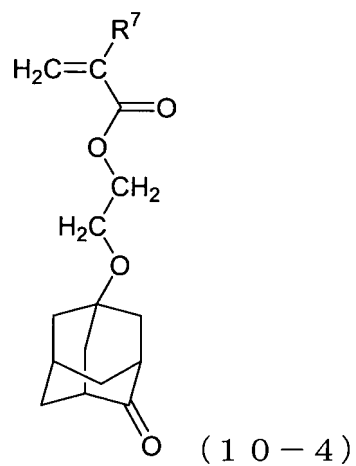
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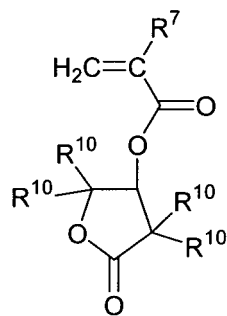
**[0095]** In the above, R7 is a hydrogen atom, methyl, trifluoromethyl or hydroxymethyl.

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**[0096]** Among the vinyl monomers of the formula (11), especially preferable are those of the formulas (11-1) to (11-15) below.

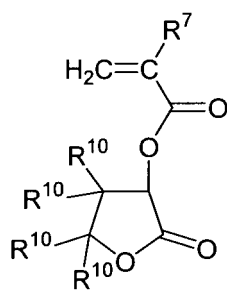
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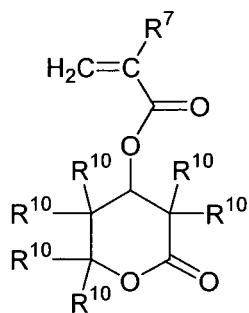
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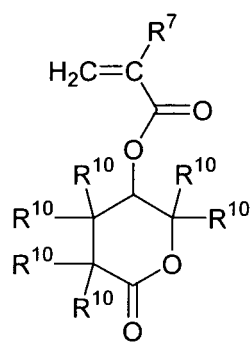
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(11-3)

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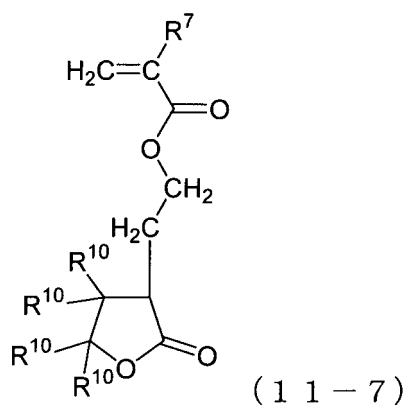
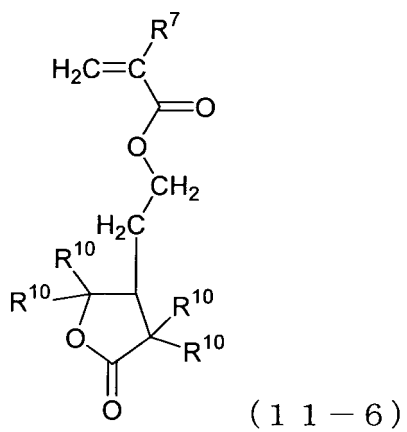
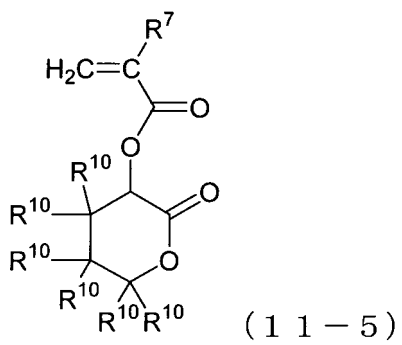


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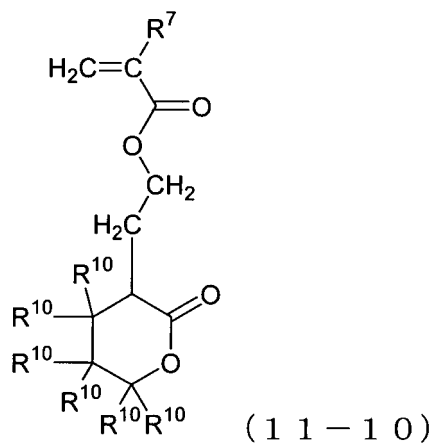
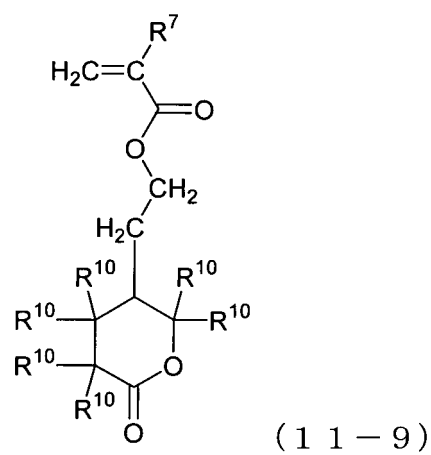
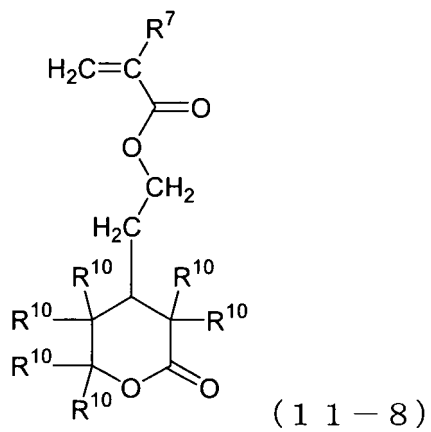
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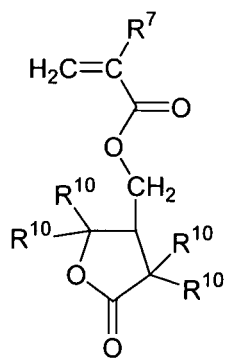
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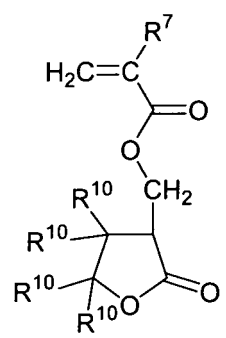


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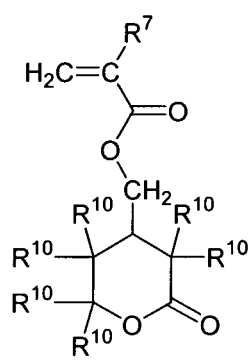
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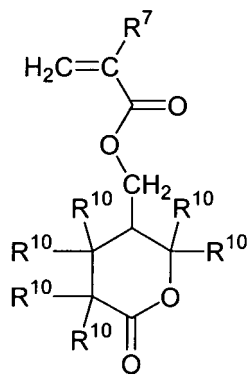
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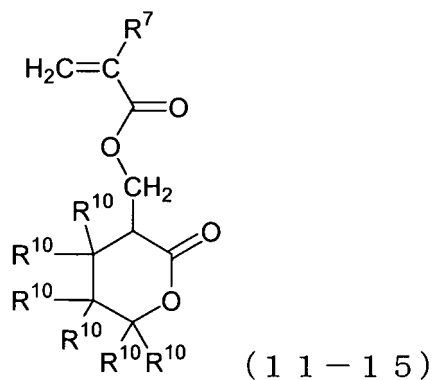
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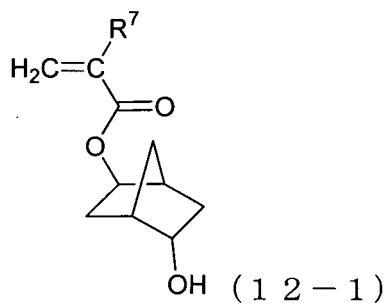
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**[0097]** In the above, preferable examples of  $R^{10}$  are a hydrogen atom, methyl, ethyl, propyl, butyl, cyclohexyl, norbornyl, adamantyl, methoxy, ethoxy, propoxy, butoxy, hydroxymethyl and hydroxyethyl.

**[0098]** Among the vinyl monomers of the formula (12), especially preferable are those of the formulas (12-1) to (12-12) below.

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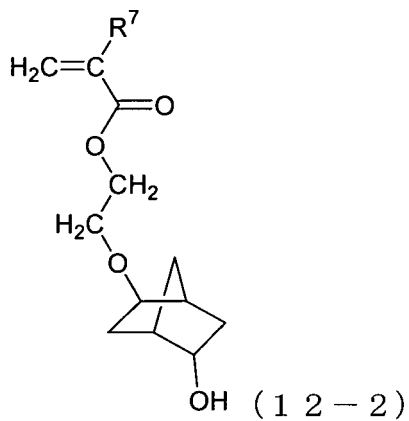


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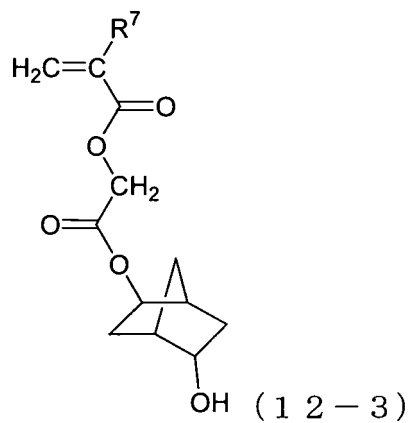


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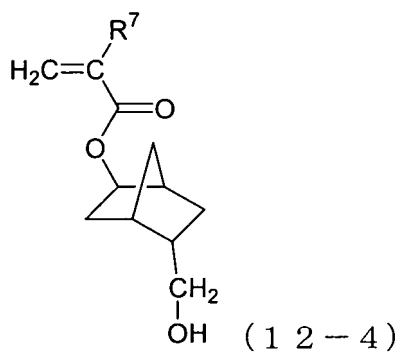
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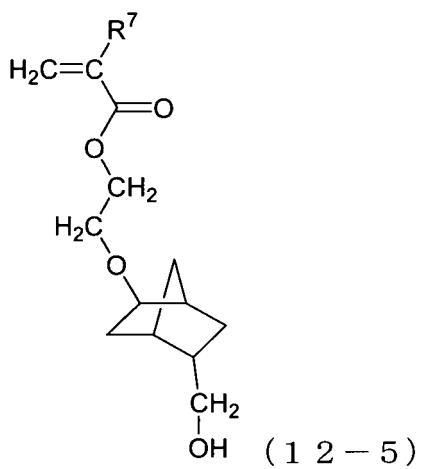
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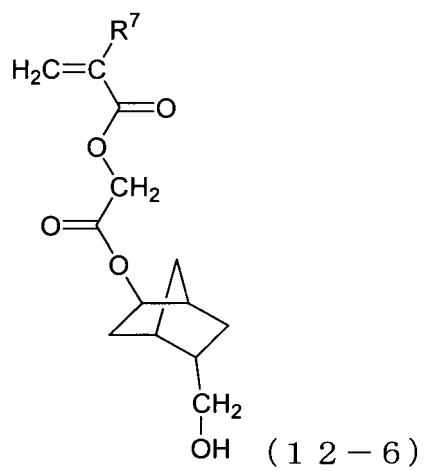
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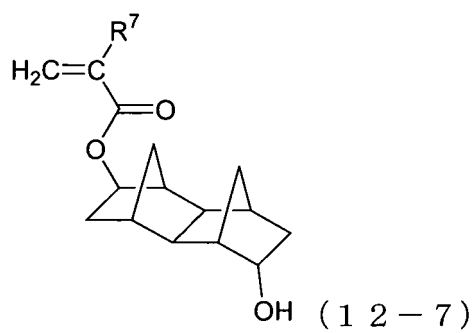
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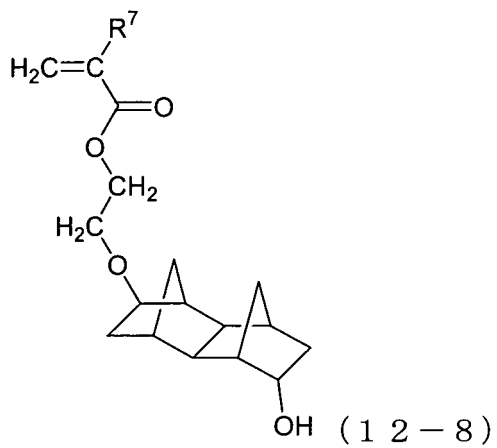
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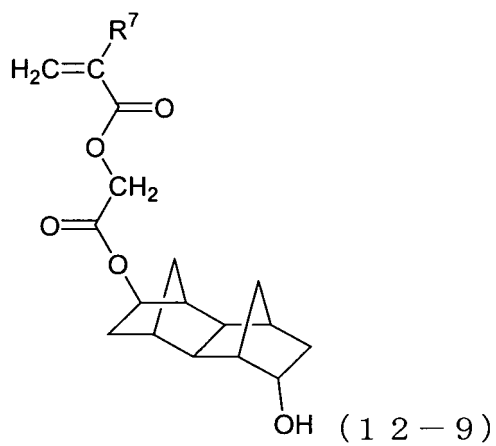
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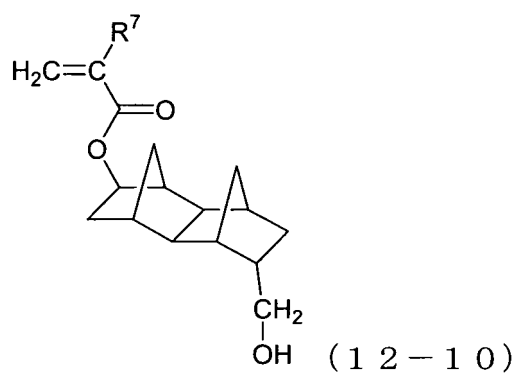
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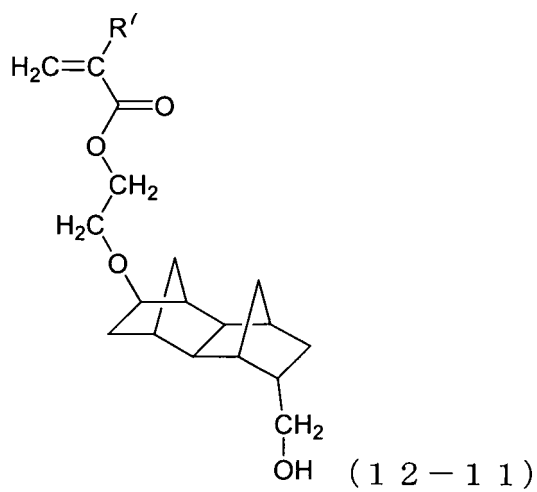
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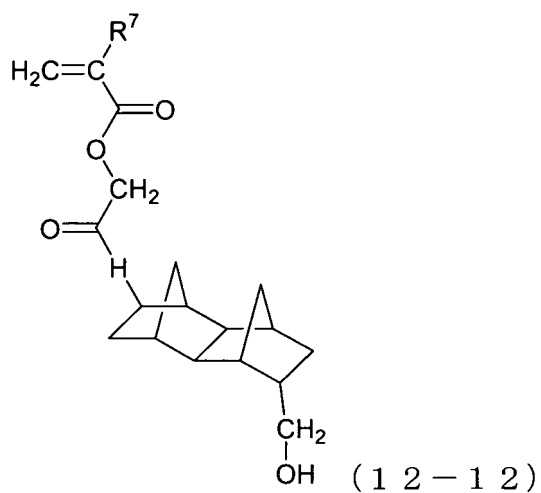
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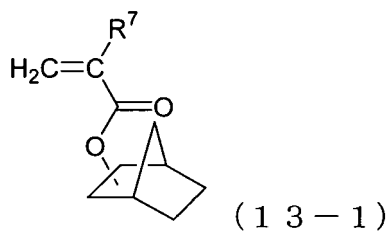


[0099] Among the vinyl monomers of the formula (13), especially preferable are those of the formulas (13-1) to (13-12) below. These monomers can be usable singly or at least two of them can be usable in mixture.

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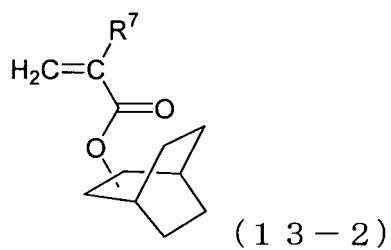
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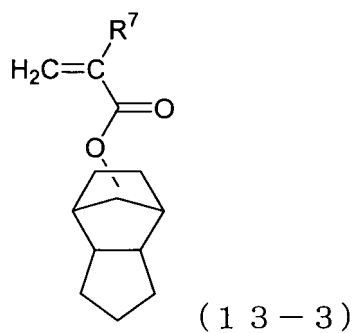
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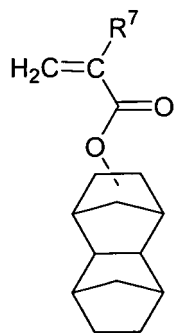
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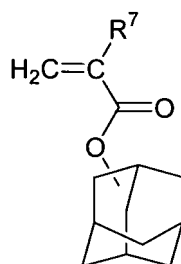
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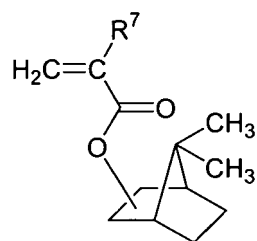
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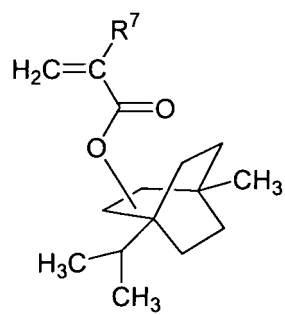
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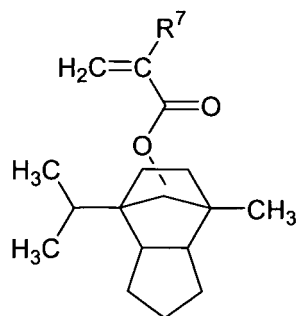
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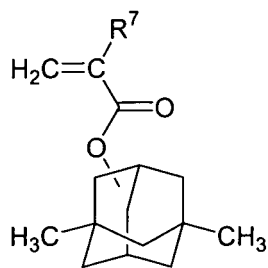
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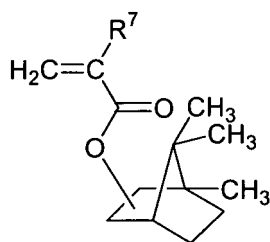
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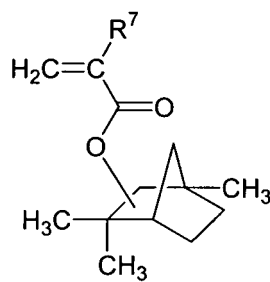
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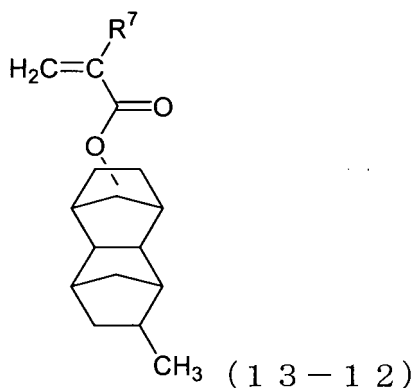
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**[0100]** The following vinyl monomers can be further usable in addition to the above monomers. Examples thereof are (meth)acrylic acid hydroxymethyl ester, 1-(meth)acrylic acid-2-hydroxymethyl ester, (meth)acrylic acid, (meth)acrylic acid-5(6)-hydroxybicyclo[2.2.1]hept-2-yl ester, (meth)acrylic acid-9(10)-hydroxytetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl, (meth)acrylic acid carboxymethyl ester, (meth)acrylic acid-2-carboxylethyl ester, (meth)acrylic acid-3-carboxyadamantane-1-yl ester, (meth)acrylic acid 5(6)-carboxybicyclo[2.2.1]hept-2-yl ester, (meth)acrylic acid-9(10)-carboxytetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl ester, (meth)acrylic cyanomethyl ester, 1-(meth)acrylic acid-2-cyanoethyl ester, (meth)acrylic acid-3-cyanoadamantane-1-yl, (meth)acrylic acid-5(6)-cyanobicyclo[2.2.1]hept-2-yl ester, (meth)acrylic acid-9(10)-cyanotetracyclo[6.2.1.1<sup>3,6</sup>.0<sup>2,7</sup>]dodeca-4-yl ester, (meth)acrylic acid methyl ester, (meth)acrylic acid ethyl ester, (meth)acrylic acid adamantane-1-yl ester, (meth)acrylic acid bicyclo[2.2.1]hept-2-yl ester, (meth)acrylic acid-7,7-dimethylbicyclo[2.2.1]hept-1-yl ester, (meth)acrylic acid tricyclo[5.2.1.0<sup>2,6</sup>]deca-8-yl ester, (meth)acrylic acid-7-oxo-6-oxa-bicyclo[3.2.1]octa-4-yl ester, (meth)acrylic acid-2-methoxycarbonyl-7-oxo-6-oxabicyclo[3.2.1]octa-4-yl ester, (meth)acrylic acid-2-oxotetrahydropyran-4-yl ester, (meth)acrylic acid-4-methyl-2-oxotetrahydropyran-4-yl ester, (meth)acrylic acid-5-oxotetrahydrofuran-3-yl ester, (meth)acrylic acid-2,2-dimethyl-5-oxotetrahydrofuran-3-yl ester, (meth)acrylic acid-4,4-dimethyl-5-oxotetrahydrofuran-3-yl ester, (meth)acrylic acid-2-oxotetrahydrofuran-3-yl ester, (meth)acrylic acid-4,4-dimethyl-2-oxotetrahydrofuran-3-yl ester, (meth)acrylic acid-5,5-dimethyl-2-oxotetrahydrofuran-3-yl ester, (meth)acrylic acid-5-oxotetrahydrofuran-2-yl methyl ester, (meth)acrylic acid-3,3-dimethyl-5-oxotetrahydrofuran-2-yl methyl ester, N, N-dimethyl (meth)acryl amide, crotonic amide, maleic amide, fumaric amide, mesaconic amide, citraconic amide, itaconic amide, etc.; methylene glycol di(meth)acrylate, ethylene glycol di(meth)acrylate, 2,5-dimethyl-2,5-hexanediol di(meth)acrylate, 1,2-adamantanediol di(meth)acrylate, 1,3-adamantanediol di(meth)acrylate, 1,4-adamantanediol di(meth)acrylate, tricyclodecanyldimethyloldi(meth)acrylate, etc.

**[0101]** The resin containing an acid-dissociable group and precision-controlled in molecular weight and molecular weight distribution ( $PD = M_w/M_n$ ) can be prepared within a short period of time in a high yield by polymerizing at least one of vinyl monomers of the formula (6) to the formula (13) using a living radical polymerization initiator of the invention represented by the formula (2).

**[0102]** Specifically stated, the living radical polymer of the present invention is produced by the process to be described below.

**[0103]** A vinyl monomer, a living radical polymerization initiator of the formula (2), and as required an azo type polymerization initiator are mixed together in a container having its inside air replaced by an inert gas. Next, the mixture is then stirred. The reaction temperature and the reaction time may be adjusted suitably. The mixture is stirred usually at 20 to 150°C for 1 minute to 100 hours, preferably at 40 to 100°C for 0.1 to 30 hours. The reaction is carried out usually under atmospheric pressure, but may be carried out at increased pressure or in a vacuum. Examples of inert gases usable at this time are nitrogen, argon, helium, etc., among which argon and nitrogen are preferred. Nitrogen is especially preferred.

**[0104]** Although the vinyl monomer and the living radical polymerization initiator of the formula (2) are used in amounts which are suitably adjusted depending on the molecular weight and molecular weight distribution of the living radical polymer to be obtained, usually 5 to 10,000 moles, preferably 50 to 5,000 moles, of the vinyl monomer is used per mole of the living radical polymerization initiator of the formula (2).

**[0105]** In case of combined use of the living radical polymerization initiator of the formula (2) and the azo type polymerization initiator, the latter is used in the ratio of usually 0.01 to 100 moles, preferably 0.1 to 10 moles, especially preferably 0.1 to 5 moles, per mole of the former. The vinyl monomer is used in an amount of 5 to 10,000 moles, preferably 50 to 5,000 moles, per mole of the living radical polymerization initiator of the formula (2).

**[0106]** The polymerization reaction is carried out usually in the absence of solvent, while an organic solvent generally in use for radical polymerization or an aqueous solvent may be used. Examples of organic solvents usable are benzene,

toluene, N,N-dimethylformamide (DMF), dimethyl sulfoxide (DMSO), acetone, 2-butanone (methyl ethyl ketone), dioxane, hexafluoroisopropanol, chloroform, carbon tetrachloride, tetrahydrofuran (THF), ethyl acetate, trifluoromethylbenzene, etc. Examples of aqueous solvents are water, methanol, ethanol, isopropanol, n-butanol, ethyl cellosolve, butyl cellosolve, 1-methoxy-2-propanol, diacetonealcohol, etc. The amount of the solvent to be used is adjusted suitably. For example,

0.01 to 50 ml, preferably 0.05 to 10 ml, more preferably 0.1 to 1 ml, of the solvent is used per gram of the vinyl monomer. **[0107]** Next, the mixture is then stirred. The reaction temperature and the reaction time may be adjusted suitably in accordance with the molecular weight or molecular weight distribution of the living radical polymer to be obtained. The mixture is stirred usually at 20 to 150°C for 1 minute to 100 hours, preferably at 40 to 100°C for 0.1 to 30 hours. The mixture is stirred more preferably at 40 to 80°C for 0.1 to 15 hours. Thus, the present invention has a feature that a high yield and precise PD are performed even at such a low polymerization temperature and short period of polymerization time. The reaction is carried out usually under atmospheric pressure, but may be carried out at increased pressure or in a vacuum.

**[0108]** After the completion of the reaction, the solvent used and the remaining monomer are removed in a vacuum to take out the desired polymer, or the desired product is isolated by re-precipitation using a solvent wherein the product is insoluble. The reaction mixture can be treated by any method insofar as it causes no problem to the desired product.

**[0109]** According to the invention, it is possible to use the emulsion polymerization process wherein a surfactant is used to carry out polymerization mainly in a micelle. When required, a dispersant of water-soluble high polymer, such as a polyvinyl alcohol, may be used. Such surfactants are usable singly, or at least two of them can be used in combination. The surfactant is used preferably in an amount of 0.3 to 50 parts by weight, more preferably 0.5 to 50 parts by weight, per 100 parts by weight of all monomers. Water is used preferably in an amount of 50 to 2000 parts by weight, more preferably 70 to 1500 parts by weight, per 100 parts by weight of all monomers. The polymerization temperature is preferably in the range of 0 to 100° C, more preferably 40 to 90° C, although not limited particularly. The reaction time is suitably determined so as to complete the polymerization reaction, for example, in accordance with the reaction temperature, or with the monomer composition to be used and the kind of surfactant and polymerization initiator. The reaction time is preferably within 24 hours.

**[0110]** According to the invention, the suspension polymerization process is also usable in which a dispersant is used to conduct polymerization generally without using any micelle. Also usable along with the dispersant are auxiliary dispersants such as sodium chloride, potassium chloride, sodium sulfate, potassium sulfate and manganese sulfate when so required. Such stabilizers for dispersion in water are used preferably in an amount of 0.01 to 30 parts by weight, more preferably 0.05 to 10 parts by weight, and most preferably 0.1 to 5 parts by weight, per 100 parts by weight of the monomers. Water is used preferably in an amount of 50 to 2000 parts by weight, more preferably 70 to 1500 parts by weight, per 100 parts by weight of the monomers. The polymerization temperature is preferably in the range of 0 to 100° C, more preferably 40 to 90° C, although not limited particularly. The reaction time is suitably determined so as to complete the polymerization reaction, for example, in accordance with the reaction temperature, or with the monomer composition to be used and the kind of water dispersion stabilizer and polymerization initiator. The reaction time is preferably within 24 hours.

**[0111]** The miniemulsion polymerization process is further usable according to the invention. The monomer is forcibly dispersed using a surfactant and cosurfactant and also using a homogenizer or ultrasonic device, but generally without using any micelle. Such surfactant and cosurfactant are used in an amount of 0.3 to 50 parts by weight, preferably 0.5 to 50 parts, based on the monomers. The ultrasonic irradiation time is 0.1 to 10 minutes, preferably 0.2 to 5 minutes.

**[0112]** According to the present invention, at least one compound selected from among a ditelluride compound, distibine compound and dibismuthine compound can be admixed with the living radical polymerization initiator of the formula (2), or with a mixture of this initiator and an azo polymerization initiator as required, for the polymerization of the vinyl monomer. Stated more specifically, the present invention provides, for example, a process for preparing a living radical polymer by polymerizing the vinyl monomer using a mixture of the living radical polymerization initiator of the formula (2) and at least one compound selected from among a ditelluride compound, distibine compound and dibismuthine compound, and a process for preparing a living radical polymer by polymerizing the vinyl monomer using a mixture of the living radical polymerization initiator of the formula (2), an azo polymerization initiator and at least one compound selected from among a ditelluride compound, distibine compound and dibismuthine compound.

**[0113]** The ditelluride compound to be used in the present invention can be a known ditelluride compound or one prepared by a common process. Examples of preparation processes are those disclosed in WO2004-014962 and WO2004-096870.

**[0114]** Examples of ditelluride compounds are dimethyl ditelluride, diethyl ditelluride, di-n-propyl ditelluride, diisopropyl ditelluride, dicyclopropyl ditelluride, di-n-butyl ditelluride, di-sec-butyl ditelluride, di-tert-butyl ditelluride, dicyclobutyl ditelluride, diphenyl ditelluride, bis(p-methoxyphenyl) ditelluride, bis(p-aminophenyl) ditelluride, bis(p-nitrophenyl) ditelluride, bis(p-cyanophenyl) ditelluride, bis(p-sulfonylphenyl) ditelluride, dinaphthyl ditelluride, dipyridyl ditelluride, etc.

**[0115]** The distibine compound to be used in the present invention can be a known distibine compound or one prepared by a common process. Examples of preparation processes are those disclosed in J. Organomet. Chem., 1973, Vol. 51,



p.223; Organometallics, 1982, Vol. 1, p.1408; and Organometallics, 1983, Vol. 2, p. 1859.

**[0116]** Examples of distibine compounds are tetramethyldistibine, tetraethyldistibine, tetraisopropyldistibine, tetrabutyl-distibine, tetravinyldistibine, tetraisopropenyldistibine, tetraisobutenyldistibine, tetraphenyldistibine, tetrakis(trimethylsilyl)distibine, 1,1'-bistibolane, tetramethyldistiboryl, etc.

**[0117]** The dibismuthine compound to be used in the present invention can be a known dibismuthine compound or one prepared by a common process. Examples of preparation processes are those disclosed in Chem. Z., 1977, Vol. 101, p. 399; and J. Organomet. Chem., 1980, Vol. 186, p. C5.

**[0118]** Examples of dibismuthine compounds are tetramethyldibismuthine, tetraethyldibismuthine, tetrapropyldibismuthine, tetraisopropyldibismuthine, tetrabutylidibismuthine, tetraisopropenyldibismuthine, tetraisobutenyldibismuthine, tetraphenyldibismuthine, tetrakis(trimethylsilyl)dibismuthine, 1,1,2,2-tetrakis(bis(trimethylsilyl)methyl)dibismuthine, 1,1'-bibismolane, etc.

**[0119]** In the case where a living radical polymer is prepared by admixing at least one compound selected from among a ditelluride compound, distibine compound and dibismuthine compound with a living radical polymerization initiator of the formula (2), or with a mixture of this initiator and an azo polymerization initiator as required, and polymerizing a vinyl monomer with use of the resulting mixture, this process can be practiced in the same manner the above-mentioned living radical polymer preparing process. Stated more specifically, the process is the same as the foregoing living radical polymer preparing process with the exception of admixing at least one compound selected from among a ditelluride compound, distibine compound and dibismuthine compound with a living radical polymerization initiator of the formula (2), or with a mixture of this initiator and an azo polymerization initiator as required, in a container in which the inside air is replaced with an inert gas.

**[0120]** The living radical polymerization initiator of the formula (2) and at least one compound selected from among a ditelluride compound, distibine compound and dibismuthine compound are used usually in such amounts that the compound selected from among a ditelluride compound, distibine compound and dibismuthine compound is used in an amount of 0.1 to 100 moles, preferably 0.1 to 10 moles, more preferably 0.1 to 5 moles, per mole of the living radical polymerization initiator of the formula (2).

**[0121]** The living radical polymerization initiator of the present invention is adapted for excellent control of molecular weights and molecular weight distributions under very mild conditions. In particular, in case of combined use of the present living radical polymerization initiator and the azo type polymerization initiator, the present polymerization reaction proceeds in a shortened reaction time than the conventional living radical polymerization reaction.

**[0122]** Different kinds of vinyl monomers are usable in the process of the invention for preparing a living radical polymer. For example when at least two kinds of vinyl monomers are reacted at the same time, a random copolymer can be obtained. The random copolymer obtained is a polymer which comprises the reacted monomers in the original ratio (mole ratio) regardless of the kinds of the monomers. When a random copolymer is obtained by reacting a vinyl monomer A and a vinyl monomer B at the same time, the copolymer has substantially the same material ratio (mole ratio). Further when two kinds of vinyl monomers are reacted in succession, a block copolymer can be obtained. The block copolymer is provided by the same order of reacted monomers regardless of the kinds of the monomers. When a vinyl monomer A and a vinyl monomer B are reacted to obtain a block copolymer, the polymer obtained is in the order of A-B or B-A in conformity with the order of monomers reacted.

**[0123]** The living radical polymer to be obtained by the invention is adjustable in molecular weight according to the reaction time and the amount of the organoantimony compound, and can be 500 to 1,000,000 in number average molecular weight. The invention is especially suitable for producing living radical polymers having a number average molecular weight of 1,000 to 50,000.

**[0124]** The living radical polymer to be obtained by the invention is controlled to 1.05 to 1.50 in molecular weight distribution ( $PD = M_w/M_n$ ). The molecular weight distribution is controllable to a narrower range of 1.05 to 1.30, a further narrower range of 1.10 to 1.20, a still narrower range of 1.09 to 1.20, 1.09 to 1.17, 1.09 to 1.12.

**[0125]** It has been found that the living radical polymer of the present invention has a growth terminal which is highly reactive organostibanyl group. Accordingly, the organoantimony compound used for radical polymerization makes it easier to convert the terminal group to other functional group than in the case of the living radical polymer obtained by conventional living radical polymerization. Further, when at least one compound selected from among ditelluride compound, distibine compound and dibismuthine compound, a growth terminal is organotellanyl group, organostibanyl group or organobismuthanyl group.

**[0126]** The living radical polymer obtained according to the invention is therefore usable as a macro living radical polymerization initiator (macroinitiator).

**[0127]** A-B diblock copolymers such as methyl methacrylate-styrene and B-A diblock copolymers such as styrene-methyl methacrylate can be obtained using a macro living radical polymerization initiator of the invention. A-B-A triblock copolymers such as methyl methacrylate-styrene-methyl methacrylate and A-B-C triblock copolymers such as methyl methacrylate-styrene-butyl acrylate are also available. This is attributable to the fact that the vinyl monomers of various different types are controllable by the living radical polymerization initiator of the formula (2), and as required the azo

type polymerization initiator, and also to the fact that highly reactive antimony is present at the growth terminal of the living radical polymer obtained with use of the living radical polymerization initiator.

**[0128]** Stated more specifically, block copolymers are prepared by the processes to be described below.

**[0129]** For preparing A-B diblock copolymers such as methyl methacrylate-styrene copolymer, methyl methacrylate, the living radical polymerization initiator of the formula (2), and as required the azo type polymerization initiator are mixed together first as in the process described above for preparing a living radical polymer to obtain poly(methyl methacrylate), and subsequently mixing styrene with the polymer to obtain methyl methacrylate-styrene copolymer.

**[0130]** A-B-A triblock copolymers and A-B-C triblock copolymers can be produced, for example, by preparing an A-B diblock copolymer by the above process and thereafter mixing a vinyl monomer (A) or vinyl monomer (C) with the copolymer to obtain the A-B-A or A-B-C triblock copolymer.

**[0131]** In producing the diblock copolymer according to the invention, the living radical polymerization initiator of the formula (2), and as required the azo type polymerization initiator and a ditelluride compound, etc. can be used when a homopolymer is prepared from the first monomer and/or when the diblock copolymer is subsequently prepared.

**[0132]** Further in producing the triblock copolymer according to the invention, the living radical polymerization initiator of the formula (2), and as required the azo type polymerization initiator and a ditelluride compound, etc. can be used at least once when a homopolymer is prepared from the first monomer, or when a diblock copolymer is subsequently prepared, or when the triblock copolymer is subsequently prepared.

**[0133]** The preparation of each block may be followed directly by the subsequent reaction for the next block, or the subsequent reaction for the next block may be initiated after the purification of the product resulting from the completion of the first reaction. The block copolymer can be isolated by a usual method. For example, the solvent used and the remaining monomer are removed in a vacuum to take out the desired polymer, or the desired product is isolated by reprecipitation using a solvent wherein the product is insoluble.

**[0134]** A resin containing an acid-dissociable group can be prepared by polymerizing a vinyl monomer using the living radical polymerization initiator of the invention and removing the growth terminal of the resulting living radical polymer. This resin is precision-controlled in molecular weight and molecular weight distribution ( $PD = M_w/M_n$ ), is highly soluble in resist solvents and can be used suitably for resists for use in manufacturing semiconductor devices which are expected to be finer in structure.

**[0135]** In the case where the living radical polymerization initiator of the formula (2) is used, a metal atom remains at the growth terminal. It is desired that the amount of the remaining metal atom be up to 25 ppm based on the whole amount of the resin to ensure improvements in resist characteristics including the sensitivity and resolution of the resist and process stability.

**[0136]** After the polymer has been formed, the metal atom remaining at the terminal of the molecule is removed by a radical reduction method using tributylstannane or a thiol compound, an adsorption method using activated carbon, silica gel, activated alumina, activated clay, molecular sieves or high polymer adsorbent, a metal adsorption method using an ion exchange resin, a liquid-liquid extraction method for removing remaining metal compounds by the combination of water for washing and a suitable solvent, a solution purification method using ultrafiltration for extracting and removing substances having a molecular weight not greater than a specified value, or a combination of such methods.

**[0137]** Given below are examples of methods of purifying the resin of the invention containing an acid-dissociable group. Metals and like impurities are removed by adsorbing the metal from a resin solution using a zeta potential filter, or by washing a resin solution with an aqueous acid solution of oxalic acid or sulfonic acid to remove the metal as chelated. The remaining monomer or oligomer component is removed to a value not greater than a specified level by a liquid-liquid extraction method for removing the remaining monomer or oligomer component by the combination of water for washing and a suitable solvent, a solution purification method using ultrafiltration for extracting and removing substances having a molecular weight not greater than a specified value, a reprecipitation method for removing the remaining monomer or the like by dropwise adding a resin solution to a poor solvent to solidify the resin, and a solid purification method wherein a resin slurry filtered off is washed with a poor solvent. Such methods can be used in combination.

**[0138]** The poor solvent to be used in the reprecipitation method is dependent, for example, on the physical properties of the resin to be purified and can not be generally exemplified. A suitable poor solvent is to be selected for use.

**[0139]** A radiation-sensitive resin composition can be obtained by using a radiation-sensitive acid producing agent, serving as a component for producing an acid when irradiated with radiation, in combination with the resin containing an acid-dissociable group.

**[0140]** The presence of the resin containing an acid-dissociable group in the radiation-sensitive resin composition renders the composition highly soluble in resist solvents, gives excellent basic properties to the resist and renders the composition extremely suitable for use in fabricating semiconductor devices which are expected to be made finer in structure.

**[0141]** Preferable examples of radiation-sensitive acid producing agents are triphenylsulfonium trifluoromethanesulfonate, triphenylsulfonium nonafluoro-n-butanesulfonate, triphenylsulfonium perfluoro-n-octanesulfonate, triphenylsul-

fonium-2-bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, triphenylsulfonium-2-(3-tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]  
 dodecanyl)-1,1-difluoroethanesulfonate, triphenylsulfonium N,N-bis(nonafluoro-n-butanesulfonyl)imidate, triphenylsul-  
 fonium camphorsulfonate, 4-cyclohexylphenyldiphenylsulfonium trifluoromethanesulfonate, 4-cyclohexylphenyldiphe-  
 5 nylsulfonium nonafluoro-n-butanesulfonate, 4-cyclohexylphenyldiphenylsulfonium perfluoro-n-octanesulfonate, 4-cy-  
 clohexylphenyldiphenylsulfonium-2-bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, 4-cyclohexylphenyldi-  
 phenylsulfonium-2-(3-tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]dodecanyl)-1,1-difluoroethanesulfonate, 4-cyclohexylphenyldiphenyl-  
 sulfonium N,N-bis(nonafluoro-n-butanesulfonyl)imidate, 4-cyclohexylphenyldiphenylsulfonium camphorsulfonate, 4-t-  
 butylphenyldiphenylsulfonium trifluoromethanesulfonate, 4-t-butylphenyldiphenylsulfonium nonafluoro-n-butanesul-  
 fonate, 4-t-butylphenyldiphenylsulfonium perfluoro-n-octanesulfonate, 4-t-butylphenyldiphenylsulfonium-2-bicyclo  
 10 [2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, 4-t-butylphenyldiphenylsulfonium-2-(3-tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]  
 dodecanyl)-1,1-difluoroethanesulfonate, 4-t-butylphenyldiphenylsulfonium-N,N-bis(nonafluoro-n-butanesulfonyl)imi-  
 date, 4-t-butylphenyldiphenylsulfonium camphorsulfonate, tri(4-t-butylphenyl)sulfonium trifluoromethanesulfonate, tri(4-t-  
 butylphenyl)sulfonium nonafluoro-n-butanesulfonate, tri(4-t-butylphenyl)sulfonium perfluoro-n-octanesulfonate, tri(4-t-  
 butylphenyl)sulfonium-2-bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, tri(4-t-butylphenyl)sulfonium 2-(3-  
 15 tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]dodecanyl)-1,1-difluoroethanesulfonate, tri(4-t-butylphenyl)sulfonium N,N-bis(nonafluoro-n-  
 butanesulfonyl)imidate, tri(4-t-butylphenyl)sulfonium camphorsulfonate, diphenyliodonium trifluoromethanesulfonate,  
 diphenyliodonium nonafluoro-n-butanesulfonate, diphenyliodonium perfluoro-n-octanesulfonate, diphenyliodonium-2-  
 bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, diphenyliodonium-2-(3-tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]dodeca-  
 nyl)-1,1-difluoroethanesulfonate, diphenyliodonium N,N-bis(nonafluoro-n-butanesulfonyl)imidate, diphenyliodonium  
 20 camphorsulfonate, bis(4-t-butylphenyl)iodonium trifluoromethanesulfonate, bis(4-t-butylphenyl)iodonium nonafluoro-n-  
 butanesulfonate, bis(4-t-butylphenyl)iodonium perfluoro-n-octanesulfonate, bis(4-t-butylphenyl)iodonium-2-bicyclo  
 [2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, bis(4-t-butylphenyl)iodonium-2-(3-tetracyclo [4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]do-  
 decanyl)-1,1-difluoroethanesulfonate, bis(4-t-butylphenyl)iodonium-N,N-bis(nonafluoro-n-butanesulfonyl)imidate, bis  
 (4-t-butylphenyl)iodonium camphorsulfonate, 1-(4-n-butoxynaphthalene-1-yl)tetrahydrothiophenium trifluorometh-  
 25 anesulfonate, 1-(4-n-butoxynaphthalene-1-yl)tetrahydrothiophenium nonafluoro-n-butanesulfonate, 1-(4-n-butoxynaph-  
 thalene-1-yl)tetrahydrothiophenium perfluoro-n-octanesulfonate, 1-(4-n-butoxynaphthalene-1-yl)tetrahydrothiophe-  
 nium-2-bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, 1-(4-n-butoxynaphthalene-1-yl)tetrahydrothiophe-  
 nium 2-(3-tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]dodecanyl)-1,1-difluoroethanesulfonate, 1-(4-n-butoxynaphthalene-1-yl)tetrahy-  
 drothiophenium-N,N-bis(nonafluoro-n-butanesulfonyl)imidate, 1-(4-n-butoxynaphthalene-1-yl)tetrahydrothiophenium  
 30 champhorsulfonate, 1-(3,5-dimethyl-4-hydroxyphenyl)tetrahydrothiophenium trifluoromethanesulfonate, 1-(3,5-dime-  
 thyl-4-hydroxyphenyl)tetrahydrothiophenium nonafluoro-n-butanesulfonate, 1-(3,5-dimethyl-4-hydroxyphenyl)tetrahy-  
 drothiophenium perfluoro-n-octanesulfonate, 1-(3,5-dimethyl-4-hydroxyphenyl)tetrahydrothiophenium-2-bicyclo[2.2.1]  
 hept-2-yl-1,1,2,2-tetrafluoroethanesulfonate, 1-(3,5-dimethyl-4-hydroxyphenyl)tetrahydrothiophenium-2-(3-tetracyclo  
 [4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]dodecanyl)-1,1-difluoroethanesulfonate, 1-(3,5-dimethyl-4-hydroxyphenyl)tetrahydrothiophenium-N,N-  
 35 bis(nonafluoro-n-butanesulfonyl)imidate, 1-(3,5-dimethyl-4-hydroxyphenyl)tetrahydrothiophenium camphorsulfonate,  
 N-(trifluoromethanesulfonyloxy)succinimide, N-(nonafluoro-n-butanesulfonyloxy)succinimide, N-(perfluoro-n-oc-  
 tanesulfonyloxy)succinimide, N-(2-bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonyloxy)succinimide, N-(2-(3-  
 tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>]dodecanyl)-1,1-difluoroethanesulfonyloxy)succinimide, N-(camphorsulfonyloxy)succinimide,  
 N-(trifluoromethanesulfonyloxy)bicyclo[2.2.1]hept-5-en-2,3-dicarboxyimide, N-(nonafluoro-n-butanesulfonyloxy)bicyclo  
 40 [2.2.1]hept-5-en-2,3-dicarboxyimide, N-(perfluoro-n-octanesulfonyloxy)bicyclo[2.2.1]hept-5-en-2,3-dicarboxyimide,  
 N-(2-bicyclo[2.2.1]hept-2-yl-1,1,2,2-tetrafluoroethanesulfonyloxy)bicyclo[2.2.1]hept-5-en-2,3-dicarboxyimide, N-(2-(3-  
 tetracyclo[4.4.0.1<sup>2,5</sup>.1.7.1<sup>0</sup>] dodecanyl) - 1,1-difluoroethanesulfonyloxy)bicyclo[2.2.1]hept-5-en-2,3-dicarboxyimide,  
 N-(camphorsulfonyloxy)bicyclo[2.2.1]hept-5-en-2,3-dicarboxyimide, etc.

**[0142]** According to the invention, radiation-sensitive acid producing agents can be used singly, or at least two of them  
 45 are usable in admixture.

**[0143]** To ensure the sensitivity of the resist and the amenability thereof to development, the radiation-sensitive acid  
 producing agent is used usually in an amount of 0.1 to 20 parts by weight, preferably 0.1 to 7 parts by weight, per 100  
 parts by weight of the resin containing an acid-dissociable group. If the amount of acid producing agent is less than 0.1  
 part by weight, lower sensitivity and impaired developability tend to result, whereas if the amount is in excess of 20 parts  
 50 by weight, lower transparency to radiation will result, making is less likely to obtain rectangular resist patterns.

#### BEST MODE OF CARRYING OUT THE INVENTION

**[0144]** The present invention will be described below in detail with reference to Reference Examples, Examples and  
 55 Test Examples, but is not limited thereto in any way.

**[0145]** The present invention will be described below in detail with reference to Examples, but is not limited thereto in  
 any way. In Examples and Comparative Examples, properties were determined by the following methods.

**[0146]** Identification of organantimony compounds and living radical polymers

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**[0147]** The organoantimony compound was identified based on the results of  $^1\text{H-NMR}$  and MS analyses. The molecular weight and molecular weight distribution of the living radical polymer were determined using GPC (gel permeation chromatography). The measuring instruments used are as follows.

$^1\text{H-NMR}$  : Varian Gemini 2000 (300MHz for  $^1\text{H}$ ), JEOL JNM-A400 (400MHz for  $^1\text{H}$ )

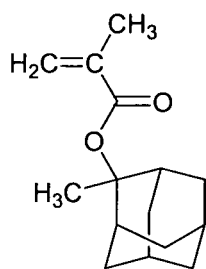
$^{13}\text{C-NMR}$  : Varian Gemini 2000, JEOL JNM-A400

MS (HRMS) : JEOL JMS-300

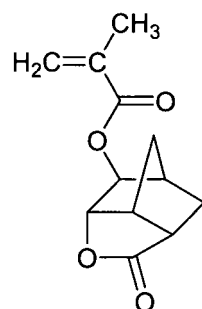
Molecular weight and molecular weight distribution: liquid chromatography Shimadzu LC-10 [column: Shodex K-804L + K-805L, polystyrene standard: TOSOH TSK Standard, poly(methyl methacrylate) standard : Shodex Standard M-75]

**[0148]** The adamantane monomer MADM and norbornene monomer NBLM used in Example 85 and Example 86 have the following structural formulae.

Adamantane monomer MADM



Norbornene monomer NBLM



### Preparation Example 1

#### Preparation of trimethylstibanyl dibromide

**[0149]** A methylmagnesium iodide solution was prepared by placing 37.7 g (1.55 moles) of magnesium and 235.4 g (1.65 moles) of methyl iodide into 900 ml of diethyl ether. A solution of 114 g (0.5 mole) of antimony trichloride in 100 ml of THF was slowly added dropwise to the iodide solution at 0° C (over a period of 40 minutes). The mixture was thereafter stirred at room temperature for 1.5 hours. A salt formed as a by-product was filtered off, and the solvent was concentrated and distilled in a vacuum (20-30° C, 200-300 mmHg). Bromine was added to the resulting liquid with stirring (until the liquid became colored owing to the addition of bromine). The precipitate obtained was washed several times with cold ether and then dried in a vacuum at room temperature, affording 115.6 g of a white solid (71% in yield).  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  revealed that the product was the desired one.

IR(KBr)3007, 1792, 1720, 1394, 874, 669, 569

$^1\text{H-NMR}$ (400MHz,  $\text{CDCl}_3$ ) 2.64(s, 9H)

$^{13}\text{C-NMR}$ (100MHz,  $\text{CDCl}_3$ ) ; 26.57

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### Preparation Example 2

#### Preparation of dimethylstibanyl bromide

- 5 **[0150]** Trimethylstibanyl dibromide (16.3 g, 50 mmole) was heated at 180° C in a vacuum (50 mmHg), and thereafter distilled, giving 9.27 g of dimethylstibanyl bromide (90.0% in yield).  
IR(neat) 2995, 2912, 1400, 1202, 1020, 843, 768, 517  
<sup>1</sup>H-NMR(400MHz, CDCl<sub>3</sub>) 1.58 (s, 6H)  
<sup>13</sup>C-NMR (100MHz, CDCl<sub>3</sub>) 8.61  
10 HRMS (EI) m/z : Calcd for C<sub>2</sub>H<sub>6</sub>BrSb(M)<sup>+</sup>, 229.8691 ; Found 229.8663.

#### Preparation Example 3 (dimethyl ditelluride)

- 15 **[0151]** A 3.19 g quantity (25 mmole) of metallic tellurium [product of Aldrich, brand name: Tellurium (-40 mesh)] was suspended in 25 ml of THF, and 25 ml (28.5 mmole) of methyl lithium [product of Kanto Chemical Co, Ltd., diethyl ether solution] was added slowly to the suspension at 0° C (10 minutes). The reaction mixture was stirred until the metallic tellurium disappeared completely (10 minutes). To the resulting reaction mixture was added 20 ml of a solution of ammonium chloride at room temperature, followed by stirring for 1 hour. The organic layer was separated off, and the aqueous layer was subjected to extraction with diethyl ether 3 times. The organic layers were collected, dried over anhydrous sodium sulfate and concentrated in a vacuum, affording 2.69 g (9.4 mmole, yield 75%) of blackish purple oil.  
20 **[0152]** The product was found to be dimethyl ditelluride by MS (HRMS) and <sup>1</sup>H-NMR.  
**[0153]** HRMS (EI) m/z: Calcd for C<sub>2</sub>H<sub>6</sub>Te<sub>2</sub>(M)<sup>+</sup>, 289.8594; Found 289.8593  
**[0154]** <sup>1</sup>H-NMR (300MHz, CDCl<sub>3</sub>) 2.67(s, 6H)

#### 25 Preparation Example 4 (tetramethyldistibine)

- [0155]** A 465 mg quantity (19.4 mmole) of magnesium (flaky) was suspended in 25 ml of THF, and a THF solution of 4.42 g (19.0 mmole) of the dimethylstibanyl bromide prepared in Preparation Example 2 was slowly added to the suspension at room temperature. The reaction mixture was thereafter stirred at 70° C for 1 hour, the solvent was concentrated in a vacuum, 20 ml of hexane was added to an oily product obtained, and the resulting solution was collected. The collected solution was concentrated in a vacuum and subsequently distilled in a vacuum (room temperature, 0.1 mmHg) to obtain 0.30 g of an oily product (1.9 mmole, 10% in yield). <sup>1</sup>H-NMR indicated that the product was tetramethyldistibine.  
30 <sup>1</sup>H-NMR(400MHz, CDCl<sub>3</sub>) 0.98(s)

#### 35 Example 1

#### Preparation of ethyl 2-dimethylstibanyl-2-methylpropionate

- 40 **[0156]** A 3.48 g quantity (30 mmole) of ethyl isobutyrate was added to 50 ml of THF, and the mixture was chilled to -78° C. To the mixture was slowly added 16.5 ml (33 mmole) of lithium isopropylamide (product of Aldrich, 2.0M heptane • THF • ethylbenzene solution) over a period of 10 minutes. The reaction mixture was stirred while being held at -78° C (for 1 hour). To the resulting solution was added 6.9 g (29.8 mmole) of dimethylstibanyl bromide at 0° C, followed by stirring at room temperature for 2 hours. After the completion of the reaction, the solvent was concentrated in a vacuum and subsequently distilled in a vacuum, giving 3.98 g of a colorless oily product (50.0% in yield), b.p. 53-55° C/1.6-1.8 mmHg. MS(HRMS), <sup>1</sup>H-NMR and <sup>13</sup>C-NMR revealed that the product was the desired one.  
45 IR(neat) 2980, 2909, 2864, 1697, 1468, 1384, 1252, 1136, 1032, 770, 515  
<sup>1</sup>H-NMR (400MHz, CDCl<sub>3</sub>) 0.76(s, 6H, SbMe<sub>2</sub>), 1.26(t, J=7.2Hz, 3H), 1.43(s, 6H), 4.11(q, J=7.2Hz, 2H)  
<sup>13</sup>C-NMR (100MHz, CDCl<sub>3</sub>) -1.51, 14.89, 23.59, 31.56, 59.84, 176.67  
50 HRMS(Cl)m/z: Calcd for C<sub>8</sub>H<sub>18</sub>O<sub>2</sub>Sb (M+H)<sup>+</sup>, 267.0345 ; Found 267.0362

#### Example 2

#### Preparation of 2-methyl-2-dimethylstibanyl-propionitrile

- 55 **[0157]** A 2.07 g (30 mmole) of isobutyronitrile was added to 50 ml of THF, and the mixture was chilled to -78° C. To the mixture was slowly added 16.5 ml (33 mmole) of lithium isopropylamide (product of Aldrich, 2.0M heptane • THF • ethylbenzene solution) over a period of 15 minutes. The reaction mixture was stirred while being held at -78° C (for 1

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hour). To the resulting solution was added 6.9 g (29.8 mmole) of dimethylstibanyl bromide at 0° C, followed by stirring at room temperature for 2 hours. After the completion of the reaction, the solvent was concentrated in a vacuum and subsequently distilled in a vacuum, giving 2.3 g of a colorless oily product (35.0% in yield), b.p. 55-57° C/3 mmHg. MS (HRMS) and <sup>1</sup>H-NMR revealed that the product was the desired one.

5 IR(Nujol) 2860, 2206, 1713, 1452, 1371, 1211, 1121, 1016, 933, 775, 721, 665, 517  
<sup>1</sup>H-NMR(400MHz, CDCl<sub>3</sub>) 0.95(s, 6H, SbMe<sub>2</sub>), 1.50(s, 6H)  
<sup>13</sup>C-NMR(100MHz, CDCl<sub>3</sub>) -0.43, 14.51, 25.22, 126.11  
 HRMS (EI) m/z : Calcd for C<sub>6</sub>H<sub>12</sub>NSb (M)<sup>+</sup>, 219.0008 ; Found 219.0028.

### 10 Example 3

(Preparation of 1-dimethylstibanyl-ethyl)benzene)

15 **[0158]** A 9.8 g quantity (30 mmole) of trimethylstibanyl bromide was suspended in 100 ml of liquid ammonia, the suspension was chilled to -78° C, and 2.87 g (125 mmole) of metal sodium was slowly added to the suspension (over a period of 10 minutes). The reaction mixture was stirred (for 1 hour) while being held at -78° C, and 2.94 g (30 mmole) of ammonium bromide was subsequently slowly added to the mixture. The resulting mixture was stirred for 1 hour, the solvent was then distilled off, and 100 ml of THF was thereafter added to the residue. To the mixture was added 6.11 g (33 mmole) of (1-bromoethyl)benzene, followed by stirring at 0° C for 4 hours. After the completion of the reaction, the solvent was concentrated in a vacuum, and the resulting mixture was distilled in a vacuum, giving 3.28 g of a colorless oily product (43% in yield), b.p. 53-57° C/1.3 mmHg. MS (HRMS), <sup>1</sup>H-NMR and <sup>12</sup>C-NMR indicated that the product was the desired one.

20 IR(neat) 2980, 2903, 2864, 1599, 1491, 1450, 1375, 1202, 1013, 764, 698, 665, 515  
<sup>1</sup>H-NMR(400MHz, CDCl<sub>3</sub>) 0.53 and 0.63(s, 6H, SbMe<sub>2</sub>), 1.60(d, J=7.2Hz, 3H), 2.99(q, J=7.2Hz, 1H), 7.02-7.06(m, 3H)  
 25 <sup>13</sup>C-NMR(100MHz, CDCl<sub>3</sub>) -2.62, -2.09, 18.79, 29.78, 124.78, 126.45, 128.35, 146.51  
 HRMS(EI)m/z: Calcd for C<sub>10</sub>H<sub>15</sub>Sb(M)<sup>+</sup>, 256.0212 ; Found 256.0207

### Example 4

30 **[0159]** Along with 1.04 g (10 mmole) of styrene, 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 was placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 100 °C for 48 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene (yield 82%).

35 **[0160]** GPC analysis (with reference to the molecular weight of an authentic sample of polystyrene) revealed Mn=7700 and PD=1.14.

### Examples 5 to 9

40 **[0161]** Along with styrene, 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) [Otsuka Chemical Co., Ltd., AIBN, hereinafter same] were placed into a glove box with the inside air replaced by nitrogen, and the reactions were carried out as shown in Table 1. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene

45 **[0162]** GPC analysis (with reference to the molecular weight of an authentic sample of polystyrene) was given in Table 1.

Table 1

	styrene (mmol)	reaction condition (°C/h)	yield (%)	Mn	PD
50	<u>Ex.5</u>	10	60/19	99	8700 1.17
	<u>Ex.6</u>	20	60/24	99	15800 1.22
	<u>Ex.7</u>	50	60/36	80	24800 1.15
	<u>Ex.8</u>	70	60/36	76	41200 1.16
55	<u>Ex.9</u>	100	60/36	65	49400 1.23

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### Example 10

5 **[0163]** Along with 1.04 g (10 mmole) of styrene, 22.0 mg (0.10 mmole) of 2-methyl-2-dimethylstibanyl-propionitrile prepared in Example 2 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 14 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 200 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene (yield 95%).

10 **[0164]** GPC analysis (with reference to the molecular weight of an authentic sample of polystyrene) revealed Mn=9000 and PD=1.17.

### Example 11

15 **[0165]** Along with 1.04 g (10 mmole) of styrene, 25.7 mg (0.10 mmole) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 16.42 mg (0.10 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 13 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 200 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene (yield 99%).

20 **[0166]** GPC analysis (with reference to the molecular weight of an authentic sample of polystyrene) revealed Mn=9300 and PD=1.22.

### Example 12

25 **[0167]** Along with 1.28 g (10 mmole) of n-butyl acrylate [stabilized with hydroquinone methyl ether (MEHQ)], 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 and 4.93 mg (0.03 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(n-butyl acrylate) (yield 96%).

30 **[0168]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=12400 and PD=1.13.

### Example 13

35 **[0169]** Along with 1.00 g (10 mmole) of methyl methacrylate [stabilized with hydroquinone (HQ)], 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 and 4.93 mg (0.03 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 4 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 100%).

40 **[0170]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=11000 and PD=1.24.

### Example 14

45 **[0171]** Along with 1.00 g (10 mmole) of methyl methacrylate (stabilized with HQ), 22.0 mg (0.10 mmole) of 2-methyl-2-dimethylstibanyl propionitrile prepared in Example 2 and 16.42 mg (0.10 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 4 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 99%).

50 **[0172]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=12700 and PD=1.25.

### Example 15

55 **[0173]** Along with 1.00 g (10 mmole) of methyl methacrylate (stabilized with HQ), 25.7 mg (0.10 mmole) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 4 hours. After the completion

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of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 92%).

**[0174]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10700 and PD=1.21.

### Example 16

**[0175]** Along with 1.13g (10 mmole) of N-isopropylacrylamide (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 12 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(N-isopropylacrylamide) (yield 99%).

**[0176]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=15600 and PD=1.07.

### Example 17

**[0177]** Along with 3.39g (50 mmole) of N-isopropylacrylamide (same as above), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 12 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(N-isopropylacrylamide) (yield 99%).

**[0178]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=34200 and PD=1.20.

### Example 18

**[0179]** Along with 0.53g (10 mmole) of acrylonitrile (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) and 1.35 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain polyacrylonitrile (yield 98%).

**[0180]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=17600 and PD=1.14.

### Example 19

**[0181]** Along with 1.11g (10 mmole) of 1-vinyl-2-pyrrolidone (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 0.5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(1-vinyl-pyrrolidone) (yield 99%).

**[0182]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10800 and PD=1.14.

### Example 20

**[0183]** Along with 3.33g (30 mmole) of 1-vinyl-2-pyrrolidone (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 7 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(1-vinyl-pyrrolidone) (yield 99%).

**[0184]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)]



revealed Mn=30600 and PD=1.21.

#### Example 21

5 **[0185]** Along with 5.55g (50 mmole) of 1-vinyl-2-pyrrolidone (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(1-vinyl-pyrrolidone) (yield 99%).

10 **[0186]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=38100 and PD=1.26.

#### Example 22

15 **[0187]** Along with 0.22g (2.5 mmole) of vinyl acetate (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 2.00 mg (0.0125 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(vinyl acetate) (yield 92%).

20 **[0188]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=2800 and PD=1.26.

#### Example 23

25 **[0189]** Along with 0.686 g (6.6 mmole) of styrene, 0.33 mg (3.3 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 15 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain a random copolymer of poly(styrene-r-methyl methacrylate) (0.94 g, yield 92.5%).

30 **[0190]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=12000 and PD=1.18.

#### Example 24

35 **[0191]** Along with 0.52 g (5.0 mmole) of styrene, 0.50 g (5.0 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 15 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain a random copolymer of poly(styrene-r-methyl methacrylate) (0.92 g, yield 90.2%).

40 **[0192]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=13500 and PD=1.16.

#### Example 25

45 **[0193]** Along with 0.343 g (3.3 mmole) of styrene, 0.66 g (6.6 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 15 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain a random copolymer of poly(styrene-r-methyl methacrylate) (0.91 g, yield 91.0%).

50 **[0194]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=12800

and PD=1.17.

#### Example 26

5 **[0195]** Along with 0.50 g (5.0 mmole) of methyl methacrylate (same as above), 0.555 g (5.0 mmole) of 1-vinyl-2-pyrrolidone, 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 27 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain a random copolymer of poly(methyl methacrylate-r-1-vinyl-2-pyrrolidone) (0.88 g, yield 81.9%).

10 **[0196]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=15200 and PD=1.23.

#### 15 Example 27

**[0197]** Along with 0.52 g (5.0 mmole) of styrene, 0.555 g (5.0 mmole) of 1-vinyl-2-pyrrolidone, 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 15 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain a random copolymer of poly(styrene-r-1-vinyl-2-pyrrolidone) (0.93 g, yield 88.2%).

20 **[0198]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=18300 and PD=1.15.

25

#### Example 28

**[0199]** Along with 1.04 g (10 mmole) of styrene and 53.4 mg (0.20 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 110 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene macroinitiator.

30 **[0200]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=4400 and PD=1.05.

35

#### Example 29

**[0201]** Along with 0.64 g (5.7 mmole) of 1-vinyl-2-pyrrolidone, 0.23 g (0.05 mmol) of polystyrene macroinitiator prepared in Example 28, 2.3 mg (0.014 mmole) of 2,2'-azobis(isobutyronitrile) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 8 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of diethyl ether which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-1-vinyl-2-pyrrolidone) .

40 **[0202]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=27100 and PD=1.05.

45

#### Example 30

**[0203]** Along with 2.00 g (20 mmole) of methyl methacrylate (same as above) and 106.8 mg (0.40 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 100 °C for 48 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) macroinitiator .

50 **[0204]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=4700 and PD=1.27.

55

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### Example 31

5 **[0205]** Along with 0.50 g (4.4 mmole) of 1-vinyl-2-pyrrolidone, 0.19 g (0.04 mmol) of poly(methyl methacrylate) macroinitiator prepared in Example 30, 1.8 mg (0.011 mmole) of 2,2'-azobis(isobutyronitrile) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 8 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(methyl methacrylate-b-1-vinyl-2-pyrrolidone).

10 **[0206]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=20500 and PD=1.30.

### Example 32

15 **[0207]** Along with 1.00 g (10 mmole) of methyl methacrylate (same as above) and 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 100 °C for 48 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 77%).

20 **[0208]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=8800 and PD=1.31.

### Example 33

25 **[0209]** Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 3.29 mg (0.02 mmol) 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 4 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 99%).

30 **[0210]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=9000 and PD=1.23.

### Example 34

35 **[0211]** Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.2 mg (0.05 mmol) 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 3.5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 92%).

40 **[0212]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10000 and PD=1.30.

### Example 35

45 **[0213]** Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 16.42 mg (0.10 mmol) 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 3.5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) (yield 94%).

50 **[0214]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=9600 and PD=1.32.

### Example 36

55 **[0215]** Along with 2.00 g (20 mmole) of methyl methacrylate (same as above), 53.4 mg (0.20 mmole) of ethyl 2-

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dimethylstibanyl-2-methylpropionate prepared in Example 1, 16.4 mg (0.10 mmol) of 2,2'-azobis(isobutyronitrile) and 2g of toluene (Wako Pure Chemical Industries, Ltd.), were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 20 hours. After the completion of the reaction, the reaction mixture was dissolved in 4 ml of tetrahydrofuran, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 100%).

**[0216]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=12,700 and PD=1.24.

### Example 37

**[0217]** In a glove box having its inside air replaced by nitrogen, a liquid mixture was prepared from 1.00 g (10 mmole) of methyl methacrylate (the same as above), 26.7 mg (0.10 mmole) of the ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 8.2 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile), 9 mg of POVAL (product of Nippon Synthetic Chemical Industry Co., Ltd., 88% in saponification degree, brand name GH-17) and 3 g of distilled water fully deaerated by nitrogen bubbling. The mixture was irradiated with ultrasonic waves for 120 seconds to obtain a uniform dispersion. The dispersion was stirred at 60° C for 20 hours. After the completion of the reaction, the reaction mixture was dissolved in 10 ml of chloroform, and the solution was placed into 300 ml of methanol with stirring. The resulting polymer precipitate was treated by suction filtration and dried, affording 94% of polymethyl methacrylate (94% in yield).

**[0218]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=9,700 and PD=1.29.

### Example 38

**[0219]** Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 30.3 mg (0.10 mmol) of tetramethyldistibine prepared in Preparation Example 4 and 3.29 mg (0.02 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 14 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 85%).

**[0220]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=8,100 and PD=1.07.

### Example 39

**[0221]** Along with 2.00 g (20 mmole) of methyl methacrylate (same as above), 53.4 mg (0.20 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 28.5 mg (0.10 mmol) of dimethyl ditelluride prepared in Preparation Example 3 and 2g of toluene (Wako Pure Chemical Industries, Ltd.), were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 90 °C for 48 hours. After the completion of the reaction, the reaction mixture was dissolved in 4 ml of tetrahydrofuran, and the solution was then poured into 300 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 95%).

**[0222]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10,600 and PD=1.08.

### Example 40

**[0223]** Along with 2.00 g (20 mmole) of methyl methacrylate (same as above), 53.4 mg (0.20 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 16.4 mg (0.10 mmol) of 2,2'-azobis(isobutyronitrile), 28.5 mg (0.10 mmol) of dimethyl ditelluride prepared in Preparation Example 3 and 2 g of toluene (Wako Pure Chemical Industries, Ltd.), were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 20 hours. After the completion of the reaction, the reaction mixture was dissolved in 4 ml of tetrahydrofuran, and the solution was then poured into 300 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 100%).

**[0224]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=12,500 and PD=1.08.

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### Example 41

5 [0225] Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 26.7 mg (0.10 mmole) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 8.2 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile), 14.3 mg (0.05 mmol) of dimethyl ditelluride prepared in Preparation Example 3 and 9 mg of POVAL (same as above), and 3 g of distilled water fully deaerated by nitrogen bubbling. The mixture was irradiated with ultrasonic waves for 120 seconds to obtain a uniform dispersion. The dispersion was stirred at 60° C for 20 hours. After the completion of the reaction, the reaction mixture was dissolved in 10 ml of chloroform, and the solution was placed into 300 ml of methanol with stirring. The resulting polymer precipitate was treated by suction filtration and dried, affording poly(methyl methacrylate) (94% in yield).

10 [0226] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10,200 and PD=1.28.

### Example 42

15 [0227] Along with 1.00 g (10 mmole) of methyl methacrylate (same as above) and 22.0 mg (0.10 mmole) of 2-methyl-dimethylstibanyl-propionitrile prepared in Example 2 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 10 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 99%).

20 [0228] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=8,300 and PD=1.50.

### Example 43

25 [0229] Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 22.0 mg (0.10 mmole) of 2-methyl-dimethylstibanyl-propionitrile prepared in Example 2 and 8.2 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 3.5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 86%).

30 [0230] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=12,100 and PD=1.37.

### Example 44

35 [0231] Along with 1.00 g (10 mmole) of methyl methacrylate (same as above) and 25.7 mg (0.10 mmole) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 100 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 84%).

40 [0232] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=9,400 and PD=1.21.

### Example 45

45 [0233] Along with 1.00 g (10 mmole) of methyl methacrylate (same as above), 25.7 mg (0.10 mmole) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 16.42 mg (0.10 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1.5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(methyl methacrylate) (yield 95%).

50 [0234] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=11,600 and PD=1.21.

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### Example 46

5 [0235] Along with 2.22 g (20.0 mmole) of 1-vinyl-2-pyrrolidone (same as above) and 106.8 mg (0.40 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 110 °C for 60 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 63%).

10 [0236] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=3,000 and PD=1.06.

### Example 47

15 [0237] Along with 0.33 g (3.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 4.10 mg (0.025 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 97%).

20 [0238] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=3,100 and PD=1.07.

### Example 48

25 [0239] Along with 1.39 g (12.5 mmole) of 1-vinyl-2-pyrrolidone (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 96%).

30 [0240] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=14,200 and PD=1.12.

### Example 49

35 [0241] Along with 2.22 g (20.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 100%).

40 [0242] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=28,300 and PD=1.09.

### Example 50

45 [0243] Along with 7.77 g (70.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 2 hours. After the completion of the reaction, the reaction mixture was dissolved in 15 ml of chloroform, and the solution was then poured into 550 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 99%).

50 [0244] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=61,300 and PD=1.27.

### Example 51

55 [0245] Along with 11.1 g (100.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 26.7 mg (0.10 mmol) of ethyl 2-

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dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 2 hours. After the completion of the reaction, the reaction mixture was dissolved in 20 ml of chloroform, and the solution was then poured into 750 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 93%).

**[0246]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=83,500 and PD=1.29.

### Example 52

**[0247]** Along with 2.0 g (18.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 17.8 mg (0.07 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 9.3 mg (0.03 mmol) of 4,4'-azobis(4-cyanovarelic acid) (Otsuka Chemical Co., Ltd., brand name: ACVA) and 2.0 g of distilled water were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, the resulting polymer precipitate was collected by suction filtration and dried to obtain poly(1-vinyl-2-pyrrolidone) (yield 100%).

**[0248]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=25,900 and PD=1.12.

### Example 53

**[0249]** Along with 1.11 g (10.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 22.0 mg (0.10 mmol) of 2-methyl 2-dimethylstibanyl-propionitrile prepared in Example 2 and 4.10 mg (0.025 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 1.5 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 94%).

**[0250]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=11,800 and PD=1.10.

### Example 54

**[0251]** Along with 1.11 g (10.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 22.0 mg (0.10 mmol) of 2-methyl 2-dimethylstibanyl-propionitrile prepared in Example 2 and 8.21 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 0.8 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 95%).

**[0252]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10,700 and PD=1.12.

### Example 55

**[0253]** Along with 1.11 g (10.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 25.7 mg (0.10 mmol) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 4.10 mg (0.025 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 4 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain of poly(1-vinyl-2-pyrrolidone) (yield 94%).

**[0254]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=11,100 and PD=1.12.

### Example 56

**[0255]** Along with 1.11 g (10.0 mmole) of 1-vinyl-2-pyrrolidone (same as above), 25.7 mg (0.10 mmol) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 8.21 mg (0.05 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 0.5 hour. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain

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of poly(1-vinyl-2-pyrrolidone) (yield 100%).

**[0256]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=12,700 and PD=1.10.

### 5 Example 57

**[0257]** Along with 1.27 g (10.0 mmole) of N-isopropylmethacrylamide (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 16.4 mg (0.10 mmol) of 2,2'-azobis(isobutyronitrile) and 3 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(N-isopropylmethacrylamide) (yield 94%).

**[0258]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=24,100 and PD=1.39.

15

### Example 58

**[0259]** Along with 0.99 g (10.0 mmole) of N,N-dimethylacrylamide (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 4.1 mg (0.025 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 20 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain poly(N,N-dimethylacrylamide) (yield 91%).

**[0260]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10,100 and PD=1.22.

25

### Example 59

**[0261]** Along with 0.99 g (10.0 mmole) of N,N-dimethylacrylamide (Wako Pure Chemical Industries, Ltd.), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 and 8.21 mg (0.50 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 20 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain poly(N,N-dimethylacrylamide) (yield 89%).

**[0262]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=8,400 and PD=1.29.

35

### Example 60

**[0263]** Along with 0.99 g (10.0 mmole) of N,N-dimethylacrylamide (Wako Pure Chemical Industries, Ltd.), 22.0 mg (0.10 mmol) of 2-methyl-2-dimethylstibanyl-propionitrile prepared in Example 2 and 4.1 mg (0.025 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 30 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain poly(N,N-dimethylacrylamide) (yield 96%).

**[0264]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10,700 and PD=1.25.

45

### Example 61

**[0265]** Along with 0.99 g (10.0 mmole) of N,N-dimethylacrylamide (Wako Pure Chemical Industries, Ltd.), 22.0 mg (0.10 mmol) of 2-methyl-2-dimethylstibanyl-propionitrile prepared in Example 2 and 8.21 mg (0.50 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 20 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain poly(N,N-dimethylacrylamide) (yield 95%).

**[0266]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=8,700 and PD=1.27.

55

### Example 62

**[0267]** Along with 0.99 g (10.0 mmole) of N,N-dimethylacrylamide (Wako Pure Chemical Industries, Ltd.), 25.7 mg



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(0.10 mmol) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 4.1 mg (0.025 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 30 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain poly(N,N-dimethylacrylamide) (yield 95%).

5 **[0268]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=11,400 and PD=1.20.

### Example 63

10 **[0269]** Along with 0.99 g (10.0 mmole) of N,N-dimethylacrylamide (Wako Pure Chemical Industries, Ltd.), 25.7 mg (0.10 mmol) of (1-dimethylstibanyl-ethyl)benzene prepared in Example 3 and 8.21 mg (0.50 mmol) of 2,2'-azobis(isobutyronitrile) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 20 hours. After the completion of the reaction, solvent and the remained monomer were removed under reduced pressure by vacuum pump to obtain poly(N,N-dimethylacrylamide) (yield 95%).

15 **[0270]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=10,200 and PD=1.24.

### Example 64

20 **[0271]** Along with 2.08 g (20.0 mmole) of styrene (same as above) and 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 110 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene macroinitiator.

25 **[0272]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=24,200 and PD=1.11.

### Example 65

30 **[0273]** Along with 0.25 g (2.2 mmole) of 1-vinyl-2-pyrrolidone, 0.20 mg (0.011 mmol) of polystyrene macroinitiator prepared in Example 64, 0.4 mg (0.003 mmol) of 2,2'-azobis(isobutyronitrile) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of diethyl ether which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-1-vinyl-2-pyrrolidone) (yield 77 %).

35 **[0274]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=50,900 and PD=1.27.

### Example 66

40 **[0275]** Along with 0.99 g (8.9 mmole) of 1-vinyl-2-pyrrolidone, 0.33 g (0.018 mmol) of polystyrene macroinitiator prepared in Example 64, 0.7 mg (0.004 mmol) of 2,2'-azobis(isobutyronitrile) and 1.5 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 21 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of diethyl ether which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-1-vinyl-2-pyrrolidone) (yield 63 %).

45 **[0276]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=74,100 and PD=1.28.

### Example 67

50 **[0277]** Along with 2.12 g (20 mmole) of 1-vinyl-2-pyrrolidone (same as above) and 106.8 g (0.40 mmol) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 100 °C for 60 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(1-vinyl-2-pyrrolidone) macroinitiator.

55 **[0278]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)]

revealed Mn=2,800 and PD=1.06.

#### Example 68

5 **[0279]** Along with 0.20 g (0.07 mmol) of poly(1-vinyl-2-pyrrolidone) macroinitiator prepared in Example 67, 1 ml of N,N-dimethylformamide (DMF), 2.9 mg (0.018 mmol) of 2,2'-azobis(isobutyronitrile) and 0.74 g (7.1 mmole) of styrene were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 8 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of diethyl ether which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(1-vinyl-2-pyrrolidone-b-styrene) (yield 14%).

10 **[0280]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=11,900 and PD=1.12.

#### Example 69

15 **[0281]** Along with 228 mg (0.08 mmol) of poly(1-vinyl-2-pyrrolidone) macroinitiator prepared in Example 67, 3.3 mg (0.02 mmol) of 2,2'-azobis(isobutyronitrile) and 1.04 g (8.1 mmole) of n-butyl acrylate (same as above) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of N,N-dimethylformamide (DMF), and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(1-vinyl-2-pyrrolidone-b-n-butyl acrylate) (yield 94%).

20 **[0282]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=19,000 and PD=1.15.

#### Example 70

25 **[0283]** Along with 134 mg (0.046 mmol) of poly(1-vinyl-2-pyrrolidone) macroinitiator prepared in Example 67, 1.9 mg (0.011 mmol) of 2,2'-azobis(isobutyronitrile) and 0.244 g (4.6 mmole) of acrylonitrile (same as above) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 8 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of N,N-dimethylformamide (DMF), and the solution was then poured into 300 ml of ether which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(1-vinyl-2-pyrrolidone-b-acrylonitrile) (yield 99%).

30 **[0284]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=34,600 and PD=1.07.

#### Example 71

35 **[0285]** Along with 2.33 g (22 mmole) of 1-vinyl-2-pyrrolidone (same as above) and 106.8 g (0.40 mmol) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 100 °C for 60 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(1-vinyl-2-pyrrolidone) macroinitiator.

40 **[0286]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=3,000 and PD=1.06.

#### Example 72

45 **[0287]** Along with 0.14 g (0.046 mmol) of poly(1-vinyl-2-pyrrolidone) macroinitiator prepared in Example 71, 1 ml of N,N-dimethylformamide (DMF), 1.9 mg (0.012 mmol) of 2,2'-azobis(isobutyronitrile) and 0.46 g (4.6 mmole) of methyl methacrylate (same as above) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 8 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of diethyl ether which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(1-vinyl-2-pyrrolidone-b-methyl methacrylate) (yield 88%).

50 **[0288]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=20,400 and PD=1.18.

## Example 73

5 [0289] Along with 183 mg (0.065 mmol) of poly(1-vinyl-2-pyrrolidone) macroinitiator prepared in Example 71, 2.7 mg (0.016 mmol) of 2,2'-azobis(isobutyronitrile) and 0.734 g (6.4 mmole) of N-isopropylacrylamide (same as above) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of N,N-dimethylformamide (DMF), and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(1-vinyl-2-pyrrolidone-b-N-isopropylacrylamide) (yield 99%).

10 [0290] GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=44,800 and PD=1.07.

## Example 74

15 [0291] Along with 1.04 g (10 mmole) of styrene (same as above) and 66.8 mg (0.25 mmol) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 110 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene macroinitiator.

20 [0292] GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=4,100 and PD=1.04.

## Example 75

25 [0293] Along with 188 mg (0.046 mmol) of polystyrene macroinitiator prepared in Example 74, 1.9 mg (0.011 mmol) of 2,2'-azobis(isobutyronitrile), 0.46 g (4.6 mmole) of methyl methacrylate (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 9 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-methyl methacrylate) (yield 89%).

30 [0294] GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=11,000 and PD=1.17.

## Example 76

35 [0295] Along with 204 mg (0.050 mmol) of polystyrene macroinitiator prepared in Example 74, 2.0 mg (0.012 mmol) of 2,2'-azobis(isobutyronitrile), 0.26 g (5.0 mmole) of acrylonitrile (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-acrylonitrile) (yield 82%).

40 [0296] GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=27,800 and PD=1.20.

## Example 77

45 [0297] Along with 0.54 g (5 mmole) of styrene (same as above) and 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 110 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain polystyrene macroinitiator.

50 [0298] GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=5,400 and PD=1.07.

## Example 78

55 [0299] Along with 264 mg (0.049 mmol) of polystyrene macroinitiator prepared in Example 77, 0.9 mg (0.005 mmol) of 2,2'-azobis(isobutyronitrile), 0.62 g (4.9 mmole) of n-butyl acrylate (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 4 hours.

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After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-n-butyl acrylate) (yield 94%).

**[0300]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=13,300 and PD=1.23.

### Example 79

**[0301]** Along with 249 mg (0.046 mmol) of polystyrene macroinitiator prepared in Example 77, 1.8 mg (0.011 mmol) of 2,2'-azobis(isobutyronitrile), 0.52 g (4.6 mmole) of N-isopropylacrylamide (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 6 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(styrene-b-N-isopropylacrylamide) (yield 99%).

**[0302]** GPC analysis [with reference to the molecular weight of an authentic sample of polystyrene] revealed Mn=34,500 and PD=1.05.

### Example 80

**[0303]** Along with 0.5 g (5 mmole) of methyl methacrylate (same as above) and 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methyl propionate prepared in Example 1 were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 110 °C for 24 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 250 ml of methanol which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(methyl methacrylate) macroinitiator.

**[0304]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=4,700 and PD=1.27.

### Example 81

**[0305]** Along with 230 mg (0.055 mmole) of poly(methyl methacrylate) macroinitiator prepared in Example 80, 0.9 mg (0.005 mmol) of 2,2'-azobis(isobutyronitrile), 0.70 g (5.5 mmole) of n-butyl acrylate (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 3 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(methyl methacrylate-b-n-butyl acrylate) (yield 97%).

**[0306]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=20,600 and PD=1.25.

### Example 82

**[0307]** Along with 66 mg (0.016 mmole) of poly(methyl methacrylate) macroinitiator prepared in Example 80, 0.6 mg (0.004 mmol) of 2,2'-azobis(isobutyronitrile), 0.18 g (1.6 mmole) of N-isopropylacrylamide (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 3 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain block copolymer of poly(methyl methacrylate-b-N-isopropylacrylamide) (yield 96%).

**[0308]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=21,600 and PD=1.15.

### Example 83

**[0309]** Along with 146 mg (0.035 mmol) of poly(methyl methacrylate) macroinitiator prepared in Example 80, 1.4 mg (0.009 mmol) of 2,2'-azobis(isobutyronitrile), 0.18 g (3.5 mmole) of acrylonitrile (same as above) and 1 ml of N,N-dimethylformamide (DMF) were placed into a glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 8 hours. After the completion of the reaction, the reaction mixture was dissolved in 5 ml of chloroform, and the solution was then poured into 300 ml of methanol which was being stirred. The resulting polymer precipitate was collected

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by suction filtration and dried to obtain block copolymer of poly(methyl methacrylate-b-acrylonitrile) (yield 62%).

**[0310]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)] revealed Mn=25,400 and PD=1.31.

### 5 Example 84

**[0311]** Along with 0.64 g (6 mmole) of N-isopropyl methacrylamide (same as above), 0.57 g (5 mmole) of N-isopropylacrylamide (same as above), 26.7 mg (0.10 mmol) of ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 16.4 mg (0.10 mmol) of 2,2'-azobis(isobutyronitrile) and 3 ml of N,N-dimethylformamide (DMF) were placed into a  
10 glove box with the inside air replaced by nitrogen, followed by stirring at 60 °C for 18 hours. After the completion of the reaction, the reaction mixture was dissolved in 3 ml of chloroform, and the solution was then poured into 200 ml of hexane which was being stirred. The resulting polymer precipitate was collected by suction filtration and dried to obtain poly(N-isopropylmethacryl amide-r-N-isopropylacrylamide) (yield 95%).

**[0312]** GPC analysis [with reference to the molecular weight of an authentic sample of poly(methyl methacrylate)]  
15 revealed Mn=24,100 and PD=1.39.

### Example 85

**[0313]** Stirred in a glove box having its inside air replaced by nitrogen at 60° C for 6 hours were 0.8 g (3.41 mmole)  
20 of an adamantane monomer MADM, 1.2 g (5.40 mmole) a norbornene monomer NBLM, 53.4 mg (0.20 mmole) of the ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 8.2 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile) and 2 g of toluene (the same as above). After the completion of the reaction, a solution of 436.5 mg (1.5 mmole) of tributyltin hydride (product of Aldrich) and 24.8 mg (0.15 mmole) of AIBN in 2 ml of toluene was added to the reaction mixture, followed by stirring at 60° C for 2 hours. After the completion of the reaction, the reaction mixture was diluted  
25 with 2 ml of THF and then placed into 300 ml of methanol with stirring. The resulting polymer precipitate was treated by suction filtration, the polymer obtained was dissolved in 10 ml of 2-butanone again, and the solution was passed through a column made of activated alumina (product of Wako Pure Chemical Ind. Ltd.). The solution was thereafter poured into 100 ml of methanol. The resulting polymer precipitate was treated by suction filtration at room temperature and dried, giving poly(MADM-r-NBLM) (1.9 g, 95% in yield).

**[0314]** GPC analysis (with reference to the molecular weight of authentic sample of polymethyl methacrylate) revealed  
30 that the resin was 7,700 in Mn and 1.38 in PD. The resin was soluble in propylene glycol monomethyl ether acetate.

### Example 86

**[0315]** Stirred in a glove box having its inside air replaced by nitrogen at 60° C for 6 hours were 0.8 g (3.41 mmole)  
35 of an adamantane monomer MADM, 1.2 g (5.40 mmole) a norbornene monomer NBLM, 53.4 mg (0.20 mmole) of the ethyl 2-dimethylstibanyl-2-methylpropionate prepared in Example 1, 8.2 mg (0.05 mmole) of 2,2'-azobis(isobutyronitrile), 57.1 mg (0.20 mmole) of the dimethyl ditelluride prepared in Preparation Example 3 and 2 g of 2-butanone. After the completion of the reaction, a solution of 523.8 mg (1.8 mmole) of tributyltin hydride (the same as above) and 29.8 mg  
40 (0.18 mmole) of AIBN in 2 ml of 2-butanone was added to the reaction mixture, followed by stirring at 60° C for 2 hours. After the completion of the reaction, the reaction mixture was diluted with 2 ml of 2-butanone and then placed into 300 ml of methanol with stirring. The resulting polymer precipitate was treated by suction filtration at room temperature, the polymer obtained was dissolved in 10 ml of 2-butanone again, and the solution was passed through a column made of activated alumina (the same as above). The solution was thereafter poured into 100 ml of methanol. The resulting  
45 polymer precipitate was treated by suction filtration at room temperature and dried, giving poly(MADM-r-NBLM) (1.84 g, 92% in yield).

**[0316]** GPC analysis (with reference to the molecular weight of authentic sample of polymethyl methacrylate) revealed  
that the resin was 8,800 in Mn and 1.21 in PD. The resin was soluble in propylene glycol monomethyl ether acetate.

### 50 INDUSTRIAL APPLICABILITY

**[0317]** The invention provides an organoantimony compound useful as a living radical polymerization initiator which realizes precision control of molecular weights and molecular weight distributions (PD=Mw/Mn) under mild conditions, processes for preparing the compound, processes for preparing polymers using the compound, and the polymers. The living radical polymers obtained by the polymerization process of the invention readily permit conversion of terminal  
55 groups to other functional groups, are useful for preparing macromonomers and useful as crosslinking sites and are usable as compatibilizing agents and as materials for block polymers.

**[0318]** A resin containing an acid-dissociable group is highly soluble in resist solvents and can be used suitably for

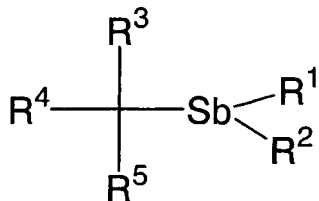
resists for use in manufacturing semiconductor devices which are expected to be finer in structure.

### Claims

5

1. An organoantimony compound represented by the formula (1)

10



(1)

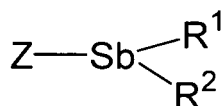
15

wherein R<sup>1</sup> and R<sup>2</sup> are C<sub>1</sub>-C<sub>8</sub> alkyl, aryl, substituted aryl or an aromatic heterocyclic group, R<sup>3</sup> and R<sup>4</sup> are each a hydrogen atom or C<sub>1</sub>-C<sub>8</sub> alkyl, and R<sup>5</sup> is aryl, substituted aryl, an aromatic heterocyclic group, oxycarbonyl or cyano.

20

2. A process for preparing an organoantimony compound of the formula (1) comprising reacting a compound of the formula (3), and a compound of the formula (4) or (5)

25

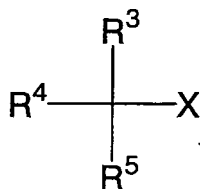


(3)

30

(wherein R<sup>1</sup> and R<sup>2</sup> are as defined above, and Z is a halogen atom or alkali metal)

35

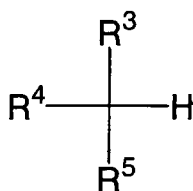


(4)

40

(wherein R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as defined above, and X is a halogen atom)

45



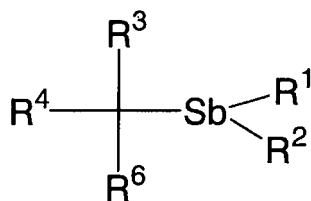
(5)

50

(wherein R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as defined above).

3. A living radical polymerization initiator of the formula (2)

55



(2)

wherein R<sup>1</sup> and R<sup>2</sup> are C<sub>1</sub>-C<sub>8</sub> alkyl, aryl, substituted aryl or an aromatic heterocyclic group, R<sup>3</sup> and R<sup>4</sup> are each a hydrogen atom or C<sub>1</sub>-C<sub>8</sub> alkyl, and R<sup>6</sup> is aryl, substituted aryl, an aromatic heterocyclic group, acyl, oxycarbonyl or cyano.

4. A process for producing a living radical polymer **characterized in that** a vinyl monomer is polymerized with use of a living radical polymerization initiator of the formula (2).
5. A process for producing a living radical polymer **characterized in that** a vinyl monomer is polymerized with use of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.
6. A living radical polymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a living radical polymerization initiator of the formula (2).
7. A living radical polymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.
8. A mixture of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.
9. A process for producing a random copolymer **characterized in that** at least two vinyl monomers are polymerized with use of a living radical polymerization initiator of the formula (2).
10. A process for producing a random copolymer **characterized in that** at least two vinyl monomers are polymerized with use of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.
11. A random copolymer which is obtainable by subjecting at least two vinyl monomers to polymerization with use of a living radical polymerization initiator of the formula (2).
12. A random copolymer which is obtainable by subjecting at least two vinyl monomers to polymerization with use of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.
13. A macro living radical polymerization initiator which is obtainable by subjecting a vinyl monomer to polymerization with use of a living radical polymerization initiator of the formula (2).
14. A macro living radical polymerization initiator which is obtainable by subjecting a vinyl monomer to polymerization with use of a living radical polymerization initiator of the formula (2) and an azo type polymerization initiator.
15. A process for producing a block copolymer **characterized in that** a vinyl monomer is polymerized with use of a macro living radical polymerization initiator of claim 13.
16. A process for producing a block copolymer **characterized in that** a vinyl monomer is polymerized with use of a macro living radical polymerization initiator of claim 14.
17. A block copolymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a macro living radical polymerization initiator of claim 13.
18. A block copolymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a macro living radical polymerization initiator of claim 14.
19. A process for producing a living radical polymer **characterized in that** a vinyl monomer is polymerized with use of

a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.

- 5 20. A process for producing a living radical polymer **characterized in that** a vinyl monomer is polymerized with use of a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 10 21. A living radical polymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 15 22. A living radical polymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 20 23. A mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 25 24. A mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 30 25. A process for producing a random copolymer **characterized in that** at least two vinyl monomers are polymerized with use of a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 35 26. A process for producing a random copolymer **characterized in that** at least two vinyl monomers are polymerized with use of a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 40 27. A random copolymer which is obtainable by subjecting at least two vinyl monomers to polymerization with use of a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 45 28. A random copolymer which is obtainable by subjecting at least two vinyl monomers to polymerization with use of a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 50 29. A macro living radical polymerization initiator which is obtainable by subjecting a vinyl monomer to polymerization with use of a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
- 55 30. A macro living radical polymerization initiator which is obtainable by subjecting a vinyl monomer to polymerization with use of a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.
31. A process for producing a block copolymer **characterized in that** a vinyl monomer is polymerized with use of a macro living radical polymerization initiator of claim 29.
32. A process for producing a block copolymer **characterized in that** a vinyl monomer is polymerized with use of a macro living radical polymerization initiator of claim 30.
33. A block copolymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a macro living radical polymerization initiator of claim 29.
34. A block copolymer which is obtainable by subjecting a vinyl monomer to polymerization with use of a macro living radical polymerization initiator of claim 30.
35. A process for producing a resin containing an acid-dissociable group **characterized in that** a vinyl monomer is



polymerized with use of one of the following (a) to (d),

- (a) a living radical polymerization initiator of the formula (2),
- (b) a mixture of a living radical polymerization initiator of the formula (2), and an azo type polymerization initiator,
- (c) a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound, and
- (d) a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.

**36.** A resin containing an acid-dissociable group which is obtainable by the process of claim 35.

**37.** A radiation-sensitive resin composition comprising a resin containing an acid-dissociable group, and a radiation-sensitive acid producing agent, the resin containing an acid-dissociable group being a resin obtained by subjecting a vinyl monomer to polymerization with use of one of the following (a) to (d),

- (a) a living radical polymerization initiator of the formula (2),
- (b) a mixture of a living radical polymerization initiator of the formula (2), and an azo type polymerization initiator,
- (c) a mixture of a living radical polymerization initiator of the formula (2), and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound, and
- (d) a mixture of a living radical polymerization initiator of the formula (2), an azo type polymerization initiator, and at least one compound selected from ditelluride compound, distibine compound and dibismuthine compound.

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP2005/012016

A. CLASSIFICATION OF SUBJECT MATTER Int.Cl. <sup>7</sup> C07F9/90, C08F4/72, 297/00, C08L57/06		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED		
Minimum documentation searched (classification system followed by classification symbols) Int.Cl. <sup>7</sup> C07F9/90, C08F4/72, 297/00, C08L57/06		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Jitsuyo Shinan Koho 1922-1996 Jitsuyo Shinan Toroku Koho 1996-2005 Kokai Jitsuyo Shinan Koho 1971-2005 Toroku Jitsuyo Shinan Koho 1994-2005		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) CAplus (STN), REGISTRY (STN)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X A	FR 1587586 A (SPOLEK PRO CHEMICKOU A HUTNI VYROBU NARODNI PODNIK), 20 March, 1970 (20.03.70), Example 5 & DE 1806388 A & NL 6815573 A	1 3-37
A	JP 43-16733 B1 (Emu Ando Tei Kemikarusu Inc.), 15 July, 1968 (15.07.68), Descriptions; page 1, left column, lines 4 to 9 (Family: none)	2
A	JP 44-3565 B1 (Emu Ando Tei Kemikarusu Inc.), 14 February, 1969 (14.02.69), Claims (Family: none)	1
<input checked="" type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents:		
"A"	document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"E"	earlier application or patent but published on or after the international filing date	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"L"	document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"O"	document referring to an oral disclosure, use, exhibition or other means	"&" document member of the same patent family
"P"	document published prior to the international filing date but later than the priority date claimed	
Date of the actual completion of the international search 08 August, 2005 (08.08.05)	Date of mailing of the international search report 23 August, 2005 (23.08.05)	
Name and mailing address of the ISA/ Japanese Patent Office	Authorized officer	
Facsimile No.	Telephone No.	

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP2005/012016

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	ANDO Fumio, et al., Insertion of Ketene and Diphenylketene to the Pnictogen-Heteroatom Bonds, Bull.Chem.Soc.Jpn., 1987, Vol.60, pages 1564 to 1566	1

Form PCT/ISA/210 (continuation of second sheet) (January 2004)

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP2005/012016

**Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)**

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1.  Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
  
2.  Claims Nos.:  
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
  
3.  Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

**Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)**

This International Searching Authority found multiple inventions in this international application, as follows:

It is recognized that a technical feature common to the inventions described in claims 1 and 2 and the inventions described in claims 3 to 37 is organic antimony compounds represented by formula (I). The compounds having such structures, however, are publicly known, as described in Example 5 in FR 1587586 A.

Accordingly, the inventions described in claims 1 and 2 and the inventions described in claims 3 to 37 are not in a technical relationship involving a common special technical feature which expressly states contribution to the prior art and thus are not found to be so linked as to form a single general inventive concept.

1.  As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2.  As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3.  As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
  
4.  No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

**Remark on Protest**

- The additional search fees were accompanied by the applicant's protest.
- No protest accompanied the payment of additional search fees.

**REFERENCES CITED IN THE DESCRIPTION**

*This list of references cited by the applicant is for the reader's convenience only. It does not form part of the European patent document. Even though great care has been taken in compiling the references, errors or omissions cannot be excluded and the EPO disclaims all liability in this regard.*

**Patent documents cited in the description**

- WO 200414848 A [0003]
- WO 2004014962 A [0113]
- WO 2004096870 A [0113]

**Non-patent literature cited in the description**

- *J. Organomet. Chem.*, 1973, vol. 51, 223 [0115]
- *Organometallics*, 1982, vol. 1, 1408 [0115]
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